

# Assessing the origin of the acidity in a humic boreal river draining peatlands and sulfide-bearing soil materials

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## Background

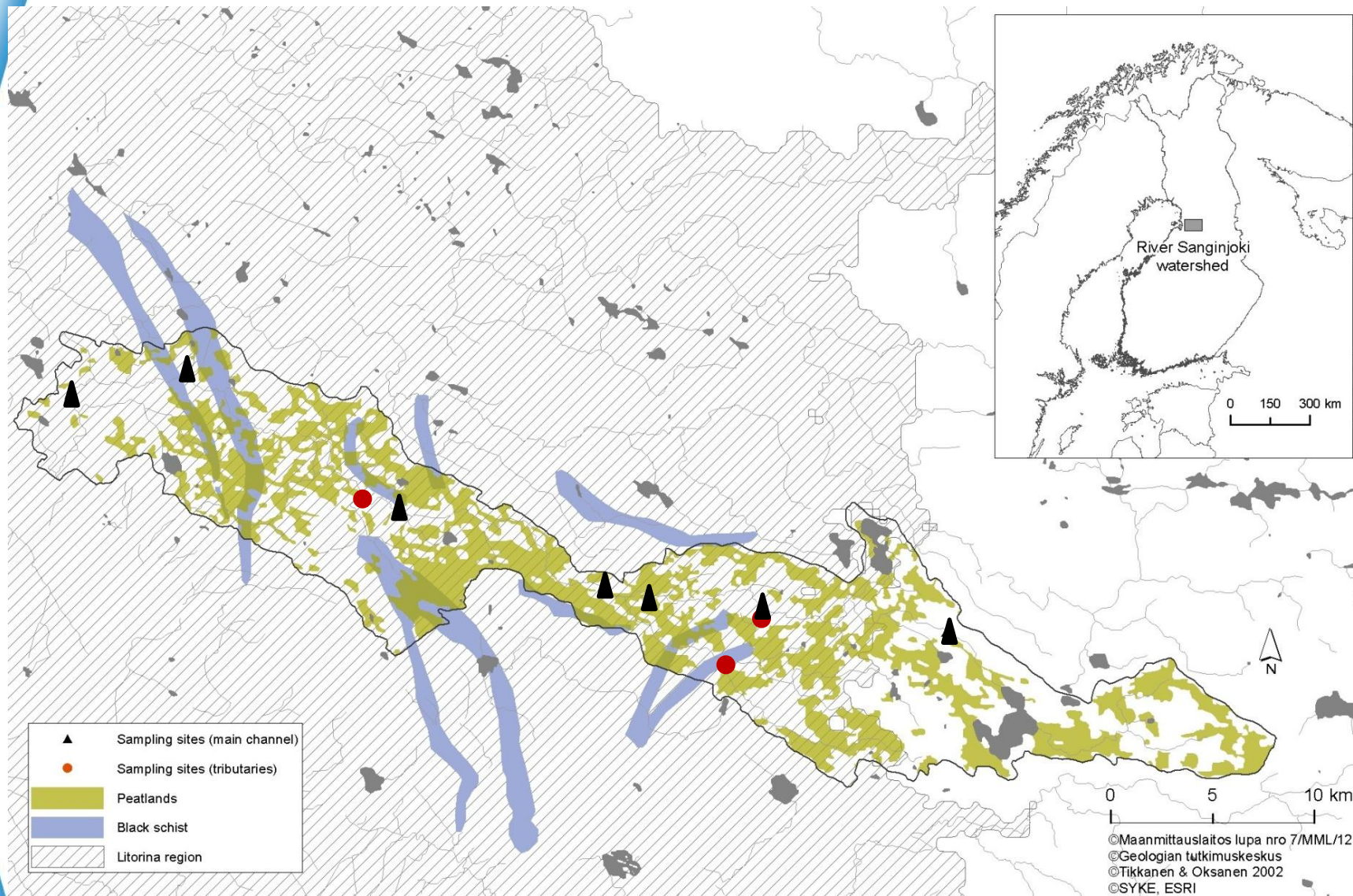
- Total catchment area is ~ 400 km<sup>2</sup> and the proportion of lakes is 2,7 %.
- Catchment area is mainly covered by mires and forests.
  - Intensively drained!
- 3/4 of the catchment area is covered by sulfide-bearing soils and black schist regions.
- Temporary acidic peaks detected during spring and autumn high runoff.
  - Fish kills in 2006-2007.

Land use	Coverage [%]
Agricultural land	1,1
Pristine peatlands	10,4
Ditched peatlands	28,7
Peat extraction	1,4
Black schist regions	8,8

### ➤ Assessment of the origin of the acidity!

- The work was conducted in City and Water – Ecological enhancement and improvement of recreational value of river Sanginjoki project!

# Catchment characteristics



- 1<sup>st</sup> sampling period: 12 set of samples in 2010 (May - October)
- 2<sup>nd</sup> period: 2 sets of samples in 2011 (April – May)

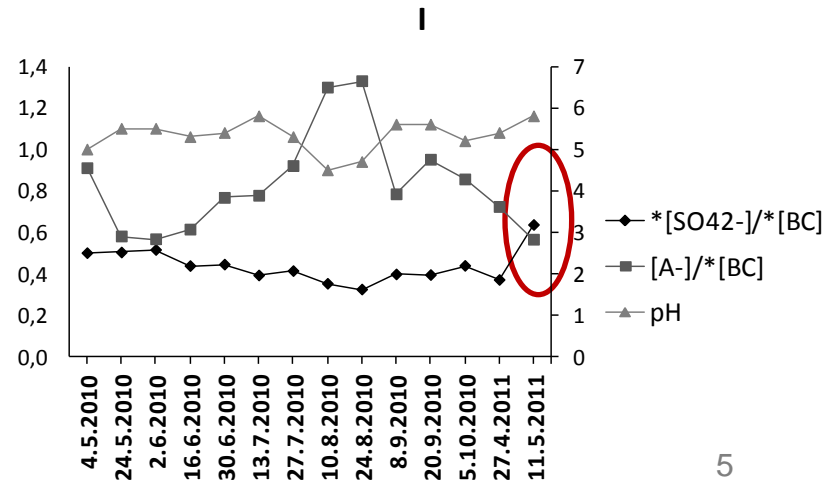
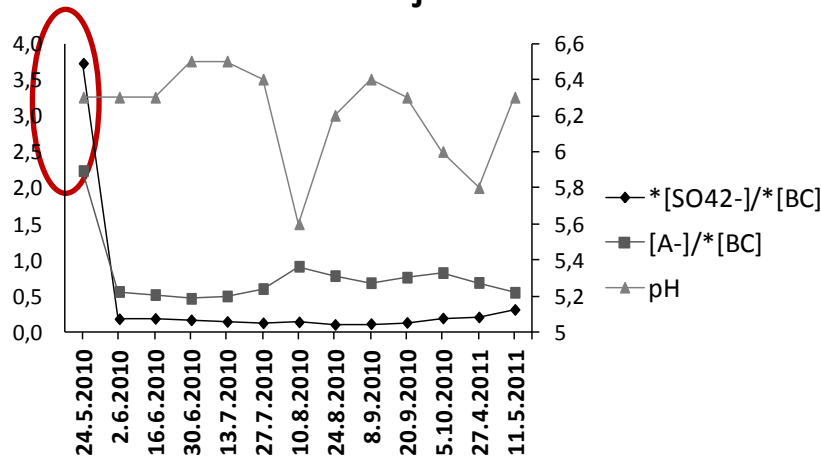
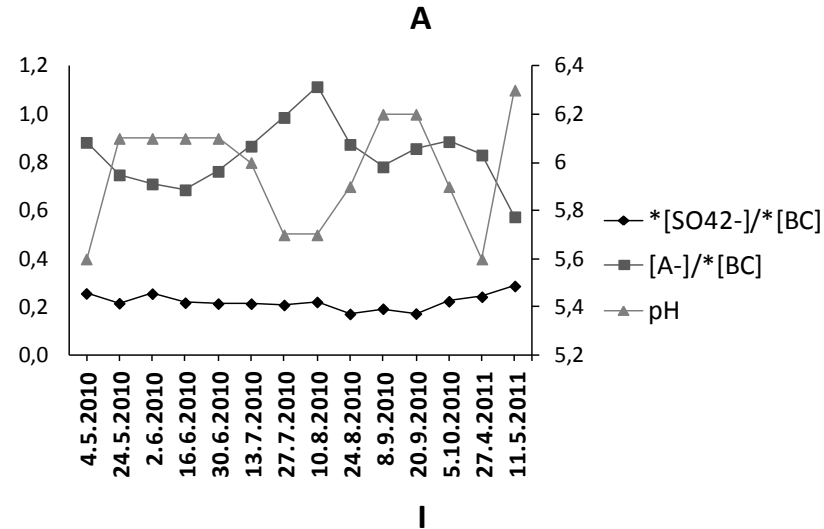
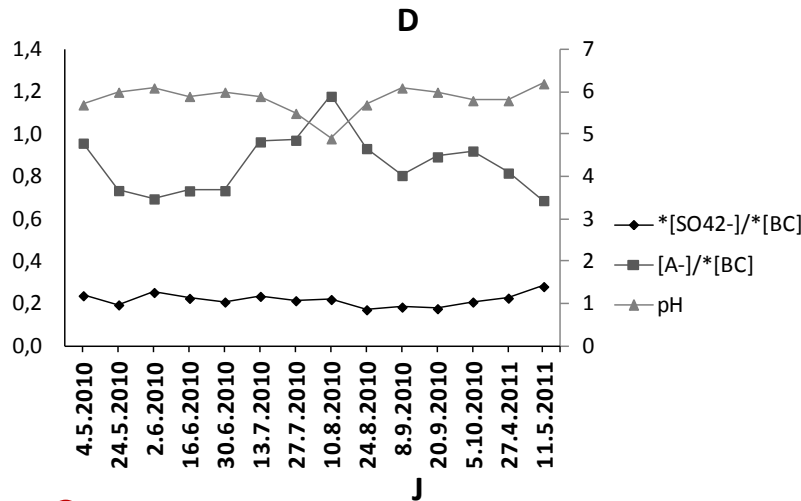
## Methods

- Due to the high coverage of peatlands and sulfide-bearing sediments/till in the watershed, the assessment of the origin of the acidity in river Sanginjoki was performed as follows:
  1. Comparison of the organic anion  $[A^-]$  and non-marine sulphate  $[*SO_4^{2-}]$  concentrations.
    - Ion balance & model based on titration results of isolated hydrophobic and hydrophilic acids<sup>(a)</sup>.
    - Sea salt corrected sulphate concentration.
  2. Statistical analysis of the water quality variables.
  3. Comparison of the collected data to a hydro-geochemistry of other streams affected by a.s. soils.

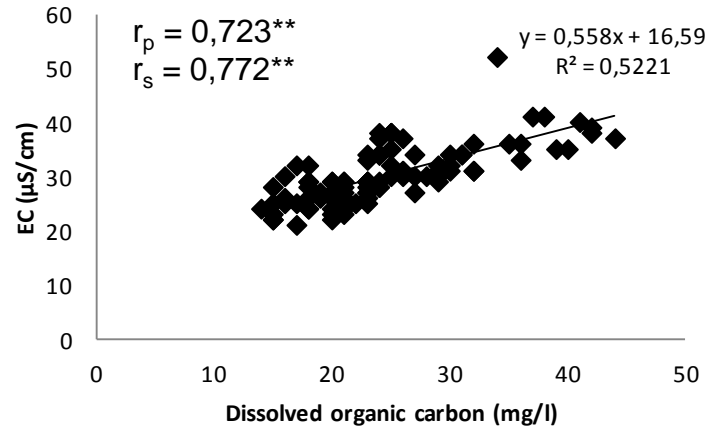
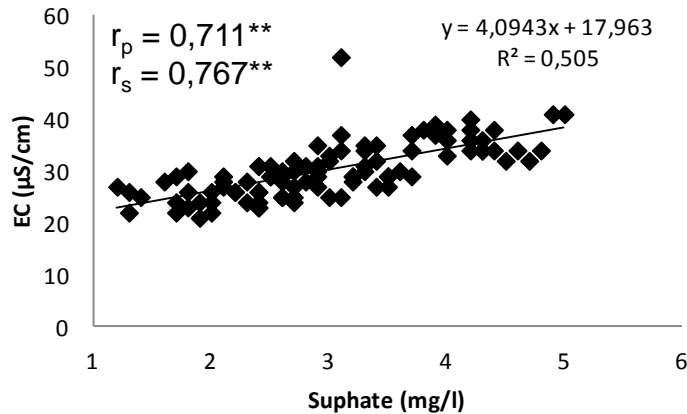
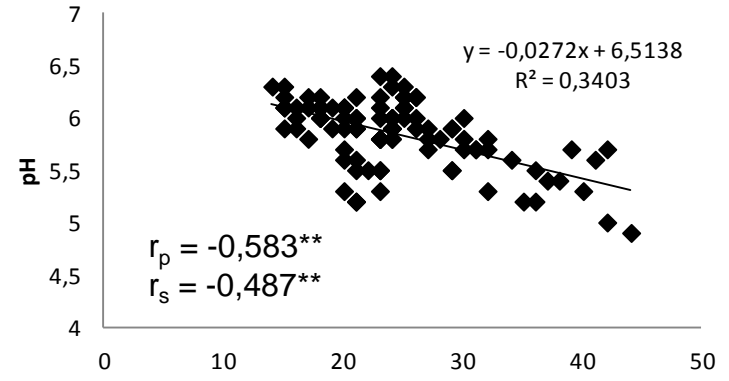
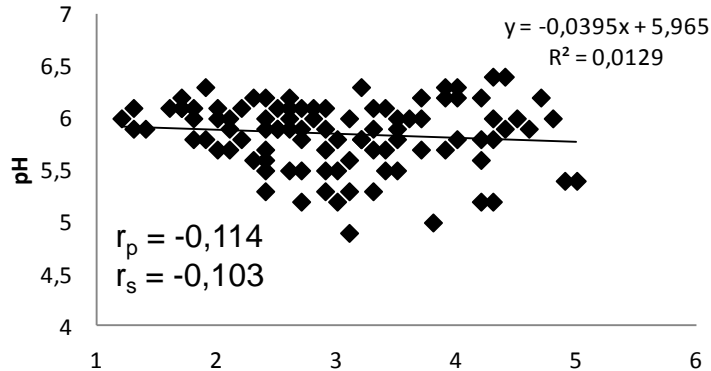
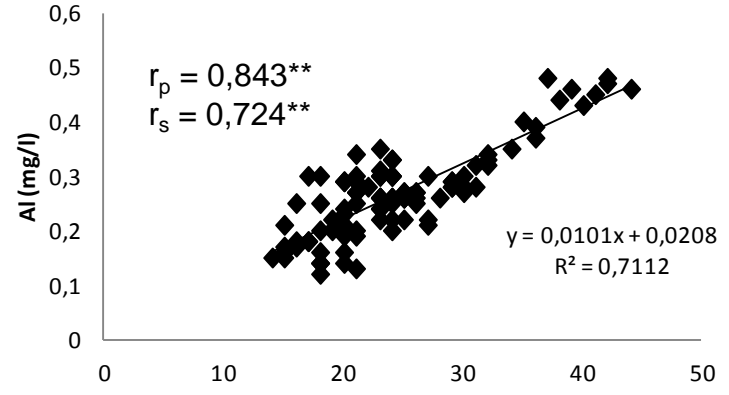
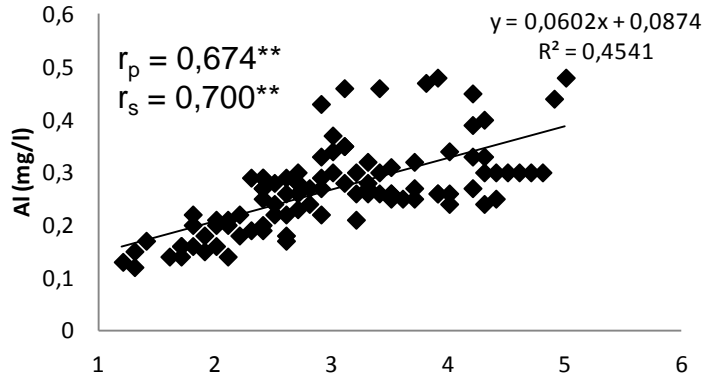
<sup>(a)</sup> Kortelainen P., Ph.D. Thesis, University of Helsinki, Publications of the Water and Environment Research Institute, no. 13, Helsinki 1993.

# Organic anion vs. non-marine sulphate

- The organic anion concentration exceeded the non-marine sulphate concentration in all main channel sampling sites.

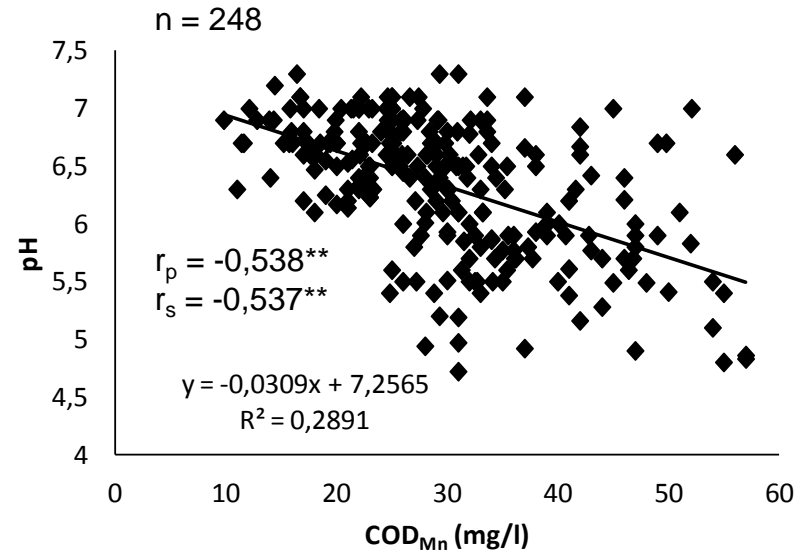
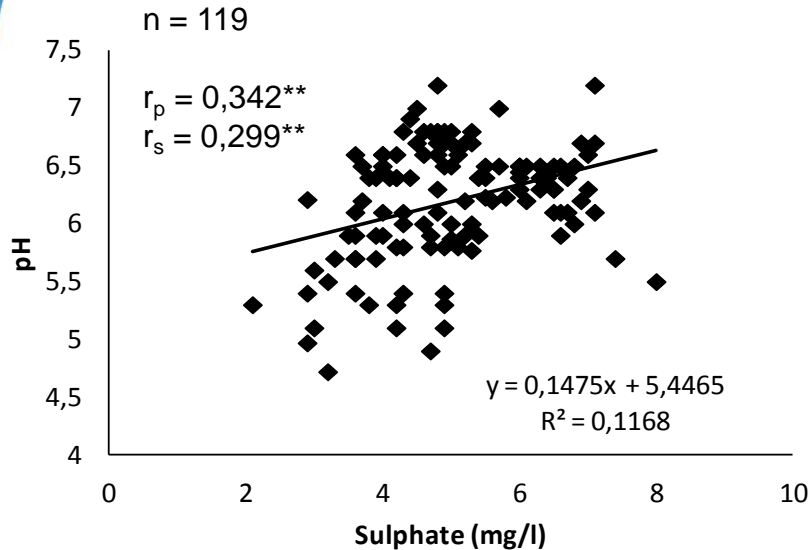


# Statistical analysis



\*\*Correlations significant at the 0,01 level (n = 98)

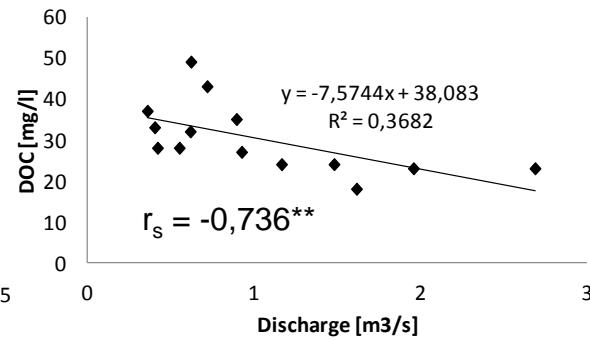
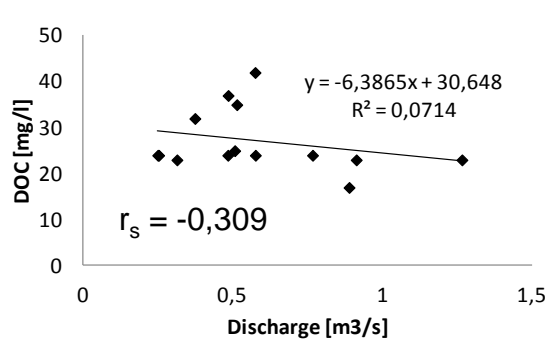
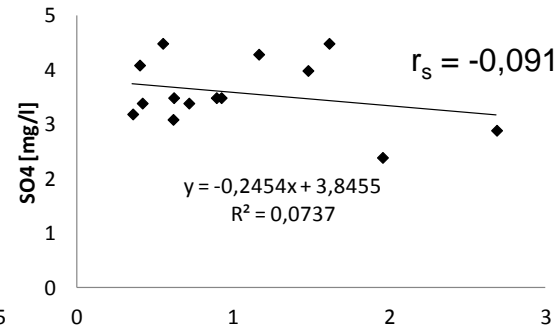
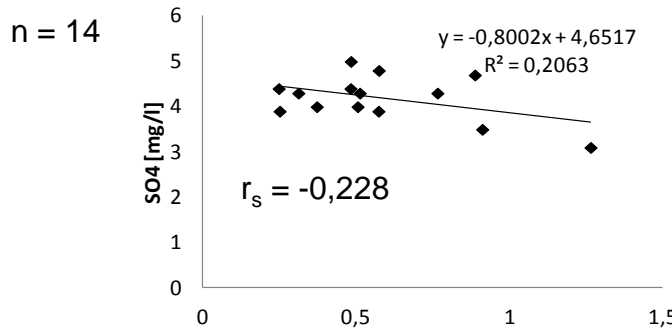
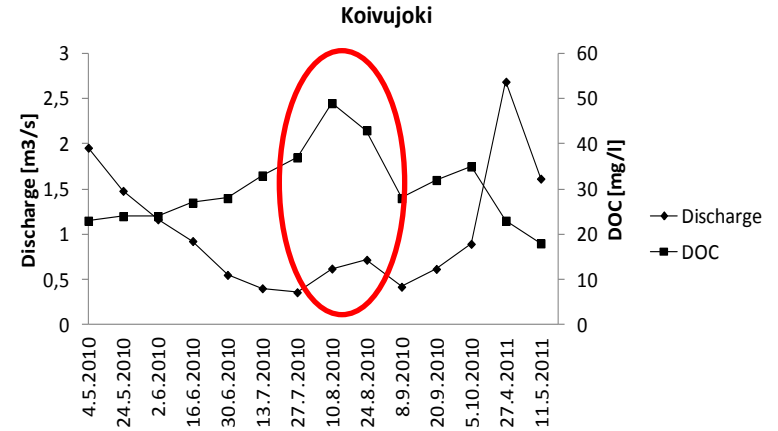
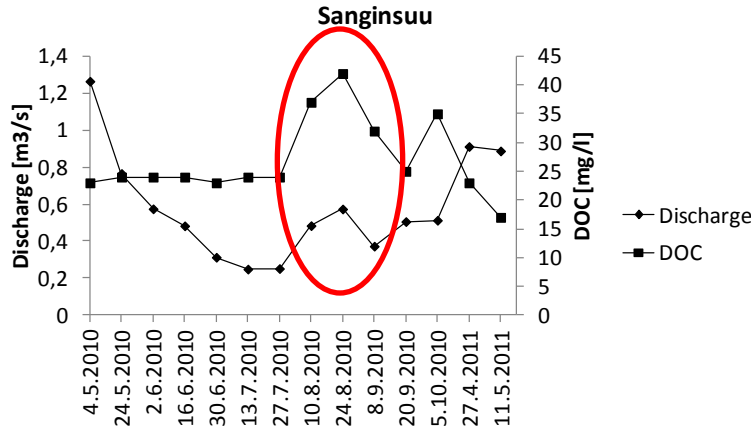
## Longer term trends



\*\*Correlations significant at the 0,01 level

- The low pH values were associated with high DOC or COD<sub>Mn</sub> concentrations in the drainage basin.
- Longer term data collected from the national Hertta database (1991-2010 for COD<sub>Mn</sub> and 1998-2001 for SO<sub>4</sub><sup>2-</sup>).

# Discharge vs. DOC/SO<sub>4</sub><sup>2-</sup>



\*\*Correlations significant at the 0,01 level



## The hydrochemistry of river Sanginjoki vs. streams affected by a.s. soils

- Detected sulphate and metal concentrations were notably higher in streams that were known to be affected by a.s. soils in the drainage basin.
- Detected EC values were also low in river Sanginjoki compared to the streams in western Finland.

	River Sanginjoki (n=139)			Streams in western Finland <sup>1</sup> (n=74)			Headwater streams in Finland <sup>2</sup> (n=1161)
	MIN	MED	MAX	MIN	MED	MAX	MED
pH	4,5	5,9	6,5	3,1	4,3	6,8	5,9
EC (µS/cm)	21	32	65	41	325	4100	44
SO <sub>4</sub> <sup>2-</sup> (mg/l)	1,2	3,2	18	6,5	85	2000	3,5
Fe (mg/l)	1,2	2,7	10	0,17	3,1	10	0,68
Al (mg/l)	0,12	0,27	0,64	0,55	5,7	267	0,09
Mn (µg/l)	6,8	51	150	BD	820	16000	29
Zn (µg/l)	2	3	9	5,4	102	2500	3,6

<sup>1</sup> Data from Åström et al., 1995

<sup>2</sup> Data from Lahermo et al., 1996.

## Conclusions

- The average organic anion concentration exceeded the non-marine sulfate concentration in all main channel sampling sites and 18 times out of 20 in the tributaries.
- Low pH values were associated with high DOC concentrations in the drainage basin.
- Detected sulfate and aluminum concentrations were low compared to other streams affected by a.s. soils.

➤ The results indicate that the leaching of organic acids from the catchment area, was the primary cause of the acidity in river Sanginjoki during sampling period!



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**Thank You!**