

Microbial Populations in Acid Sulfate Soils: Potential Role in Metal and Acid Release

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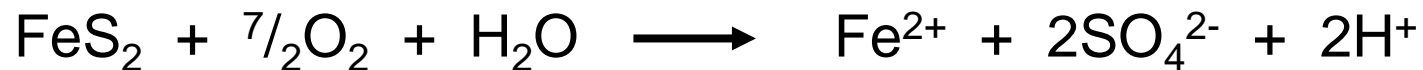
Lecture

- Metal dissolution & acid generation
- Microorganisms in acid environments
- Microorganisms in PASS & ASS
- Conclusions



Microbial Facilitated Pyrite Dissolution

Chemical oxidation:



Oxidation by Ferric Iron:



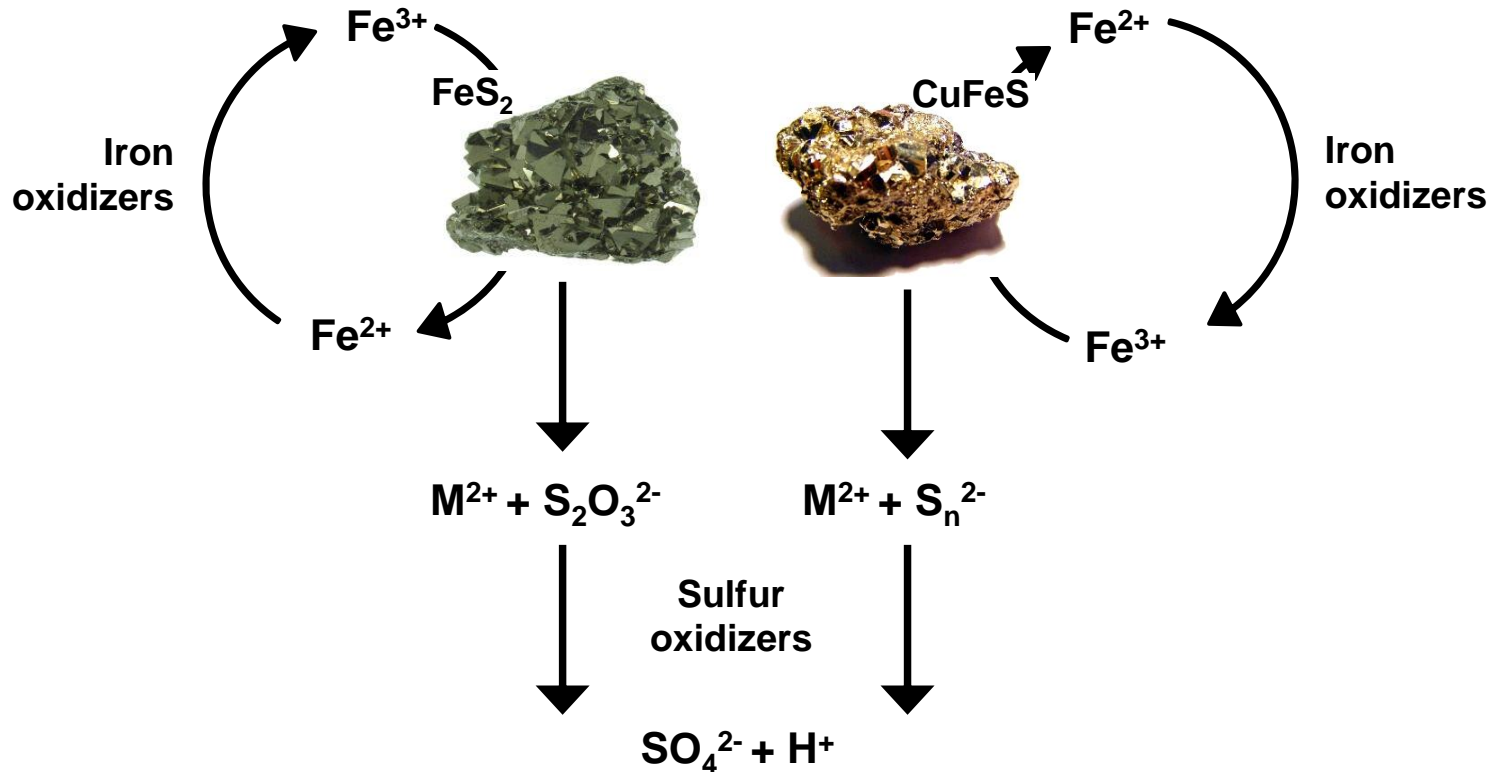
Fe³⁺ provided by microbes:



Metal Dissolution & Acid Generation

Thiosulfate mechanism

Polysulfide mechanism



Modified from Schippers & Sand (1999) Appl Environ Microbiol 65: 319-321



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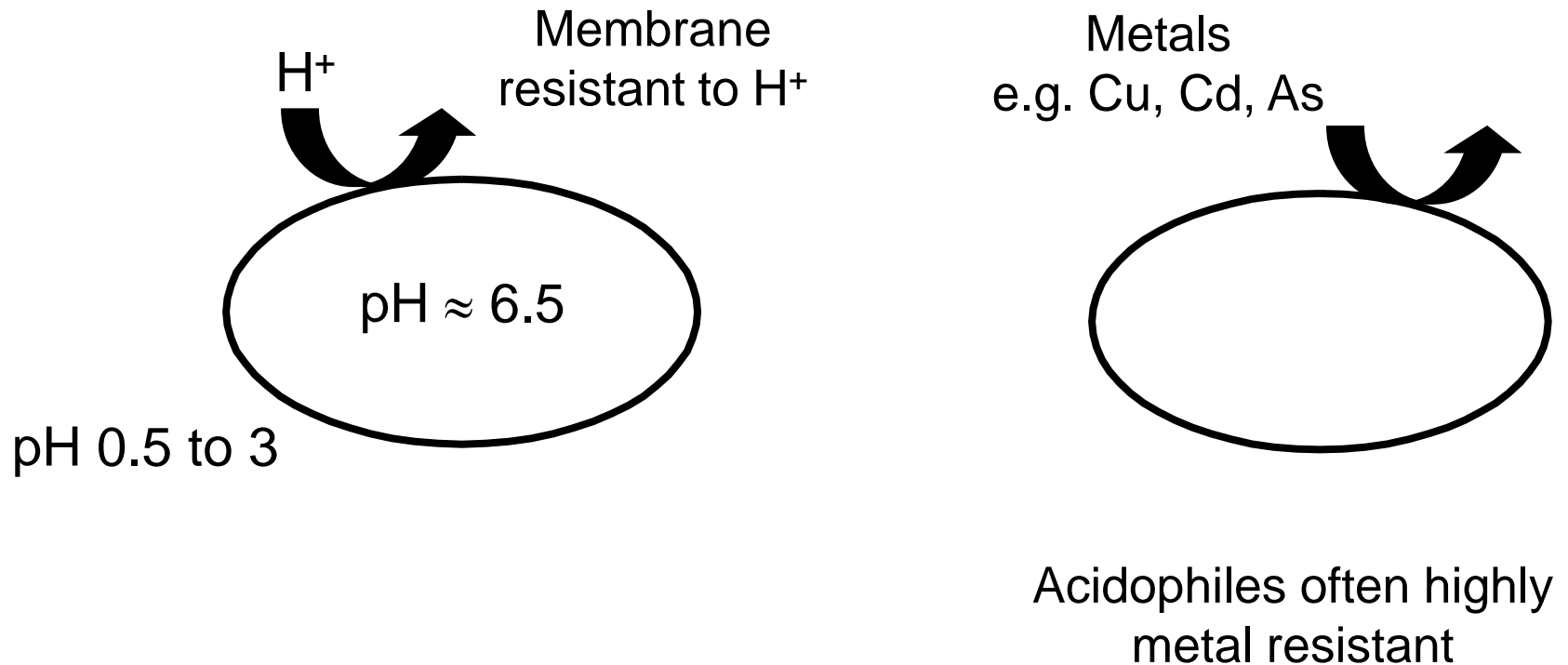


Microorganisms in Acid Environments

- Found in all domains of life
- Heterotrophs (utilize organic carbon) & autotrophs (use CO_2 from the air)
- Iron & sulfur compound oxidizers
- Adapted to low pH and high metal concentrations



Microorganisms in Acid Environments



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Microorganisms in PASS & ASS

Plough layer; pH >4.7
(30cm below the surface) →

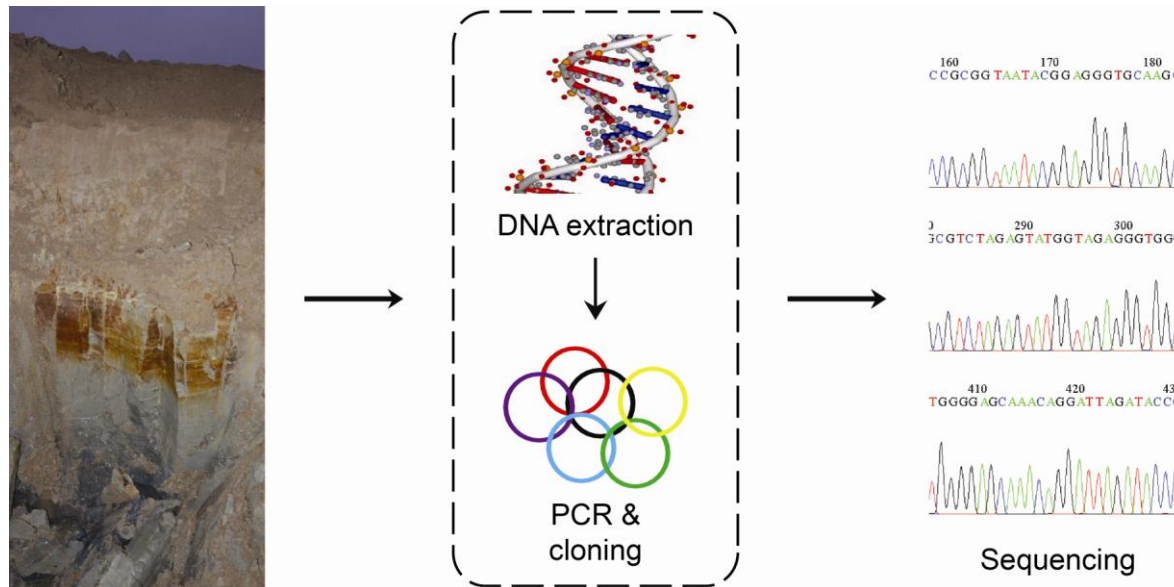
Red oxidized; pH 3.7 - 4.2 (75 cm) →

Mixed partially oxidized; pH 4 to 6 (127 cm) →

Dark reduced zones; pH ~ 7 (>180 cm) →

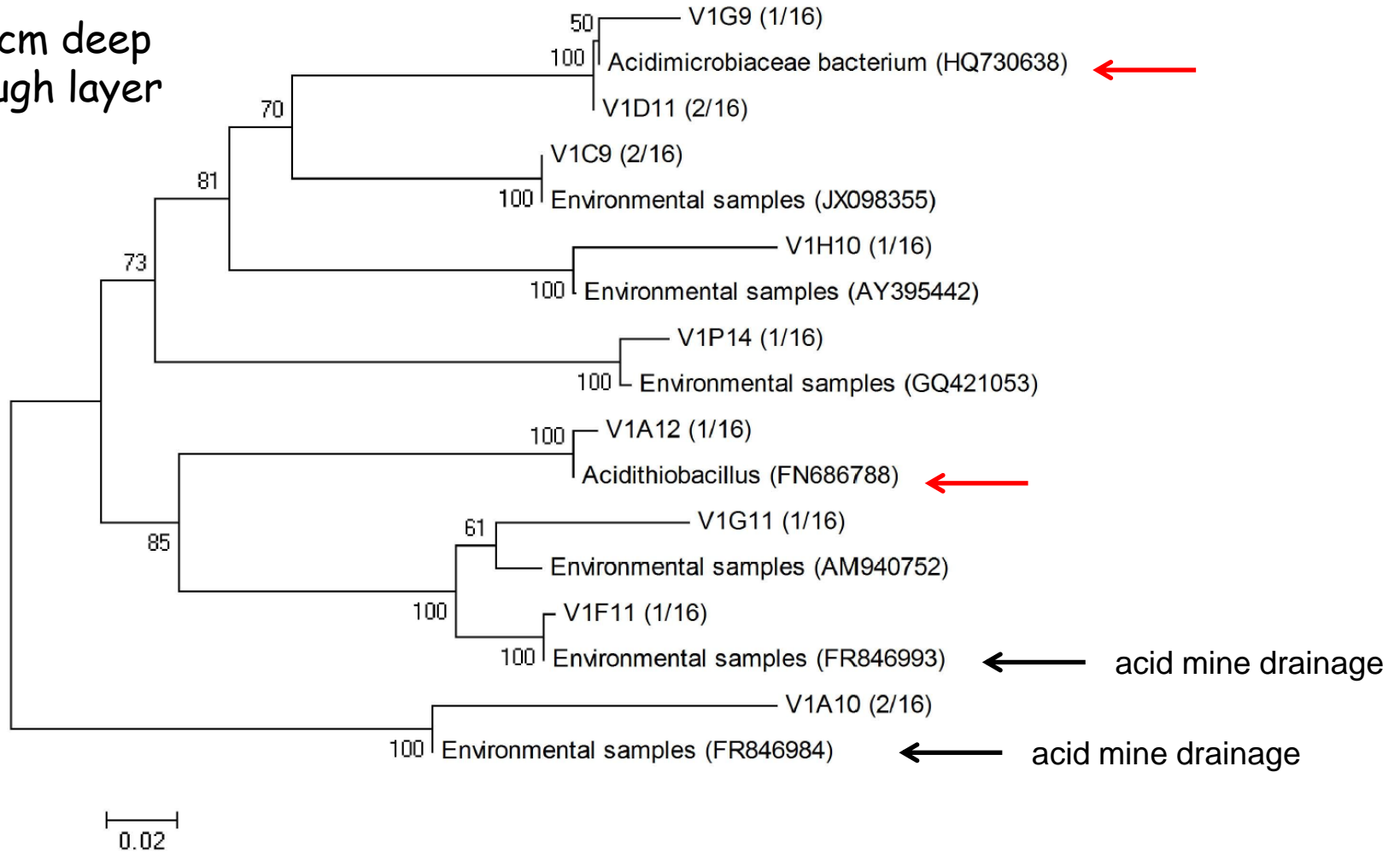


Microbial Identification

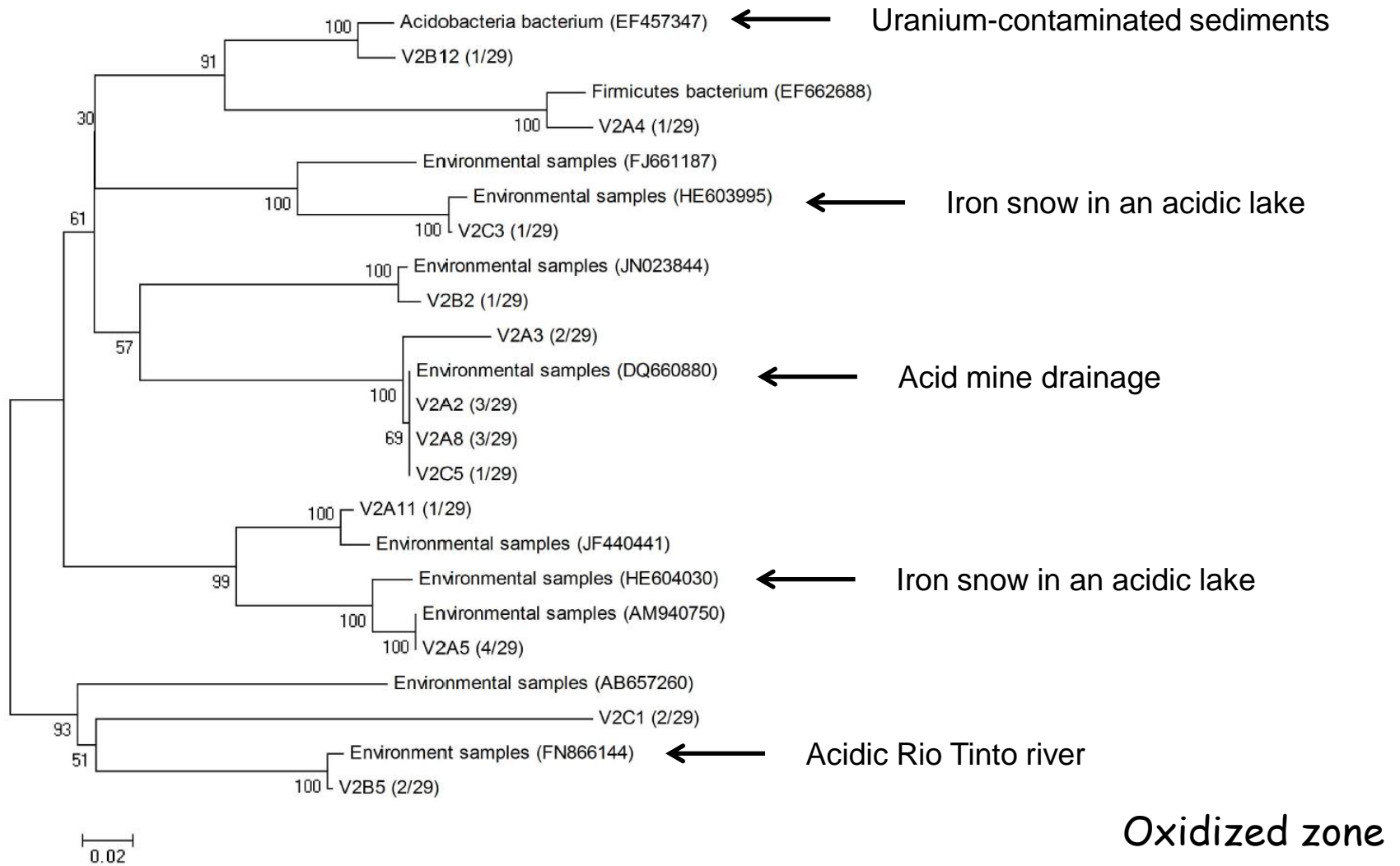


Microorganisms in PASS & ASS

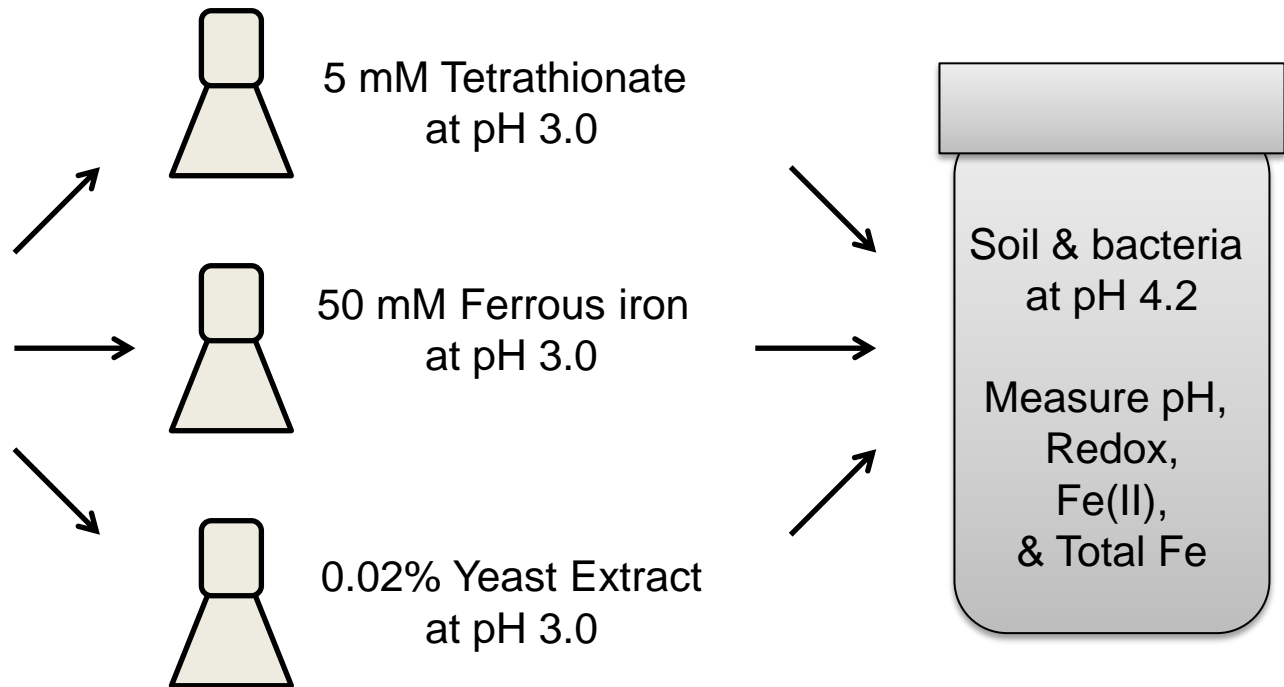
30 cm deep
plough layer



Microorganisms in PASS & ASS

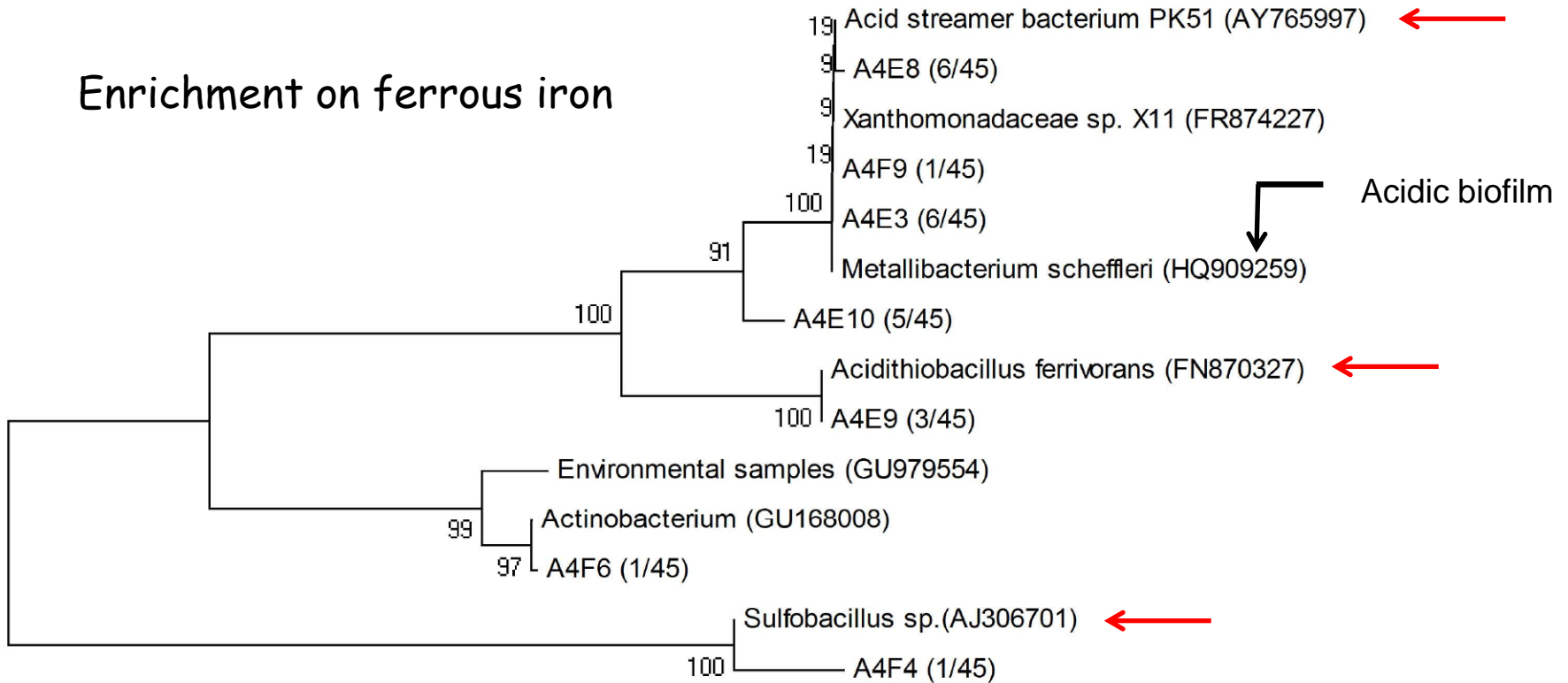


Metal & Acid Release



Enriched Microorganisms

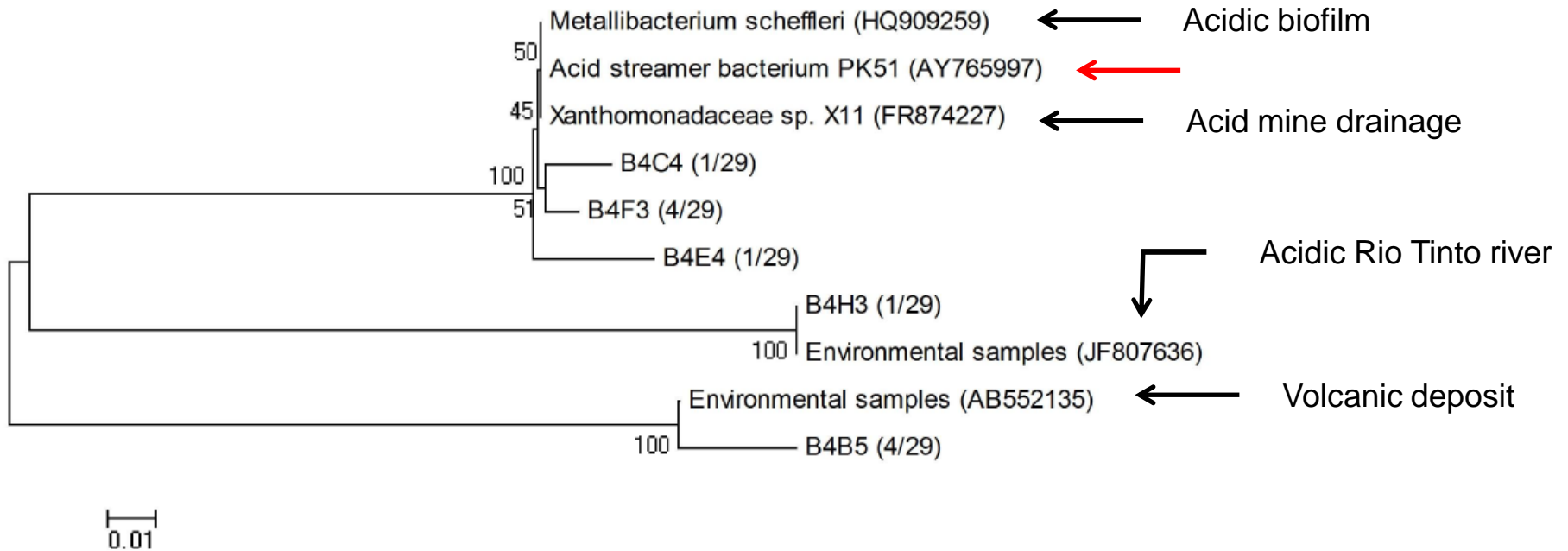
Enrichment on ferrous iron



0.02



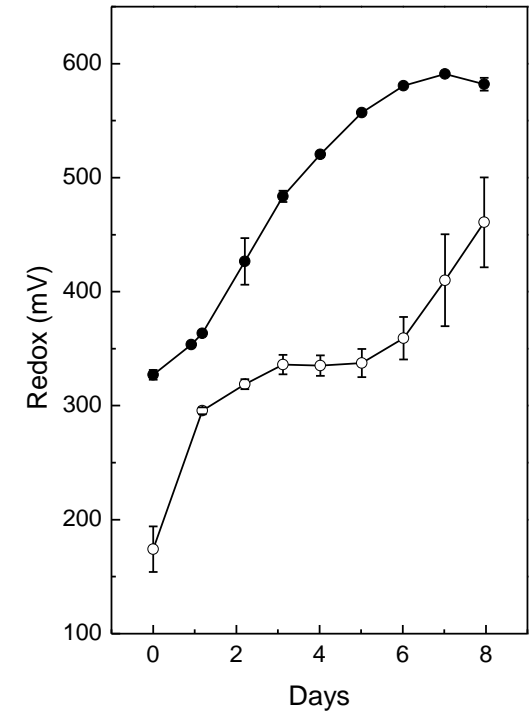
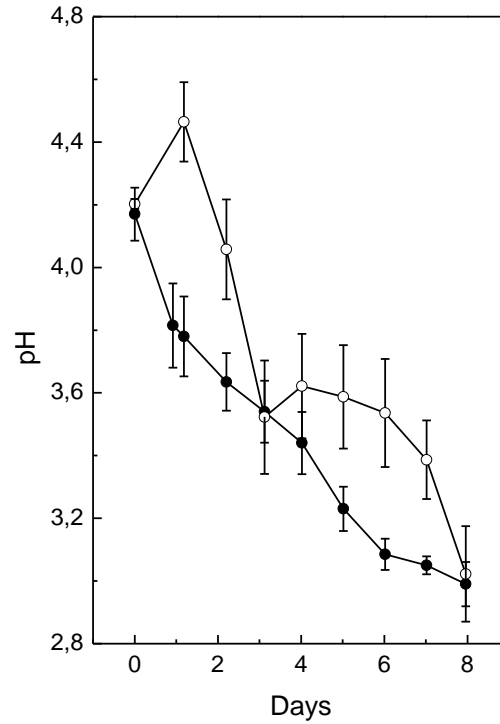
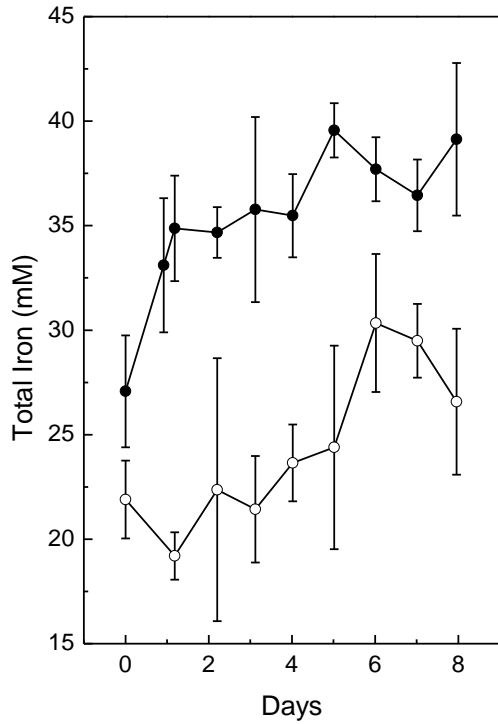
Enriched Microorganisms



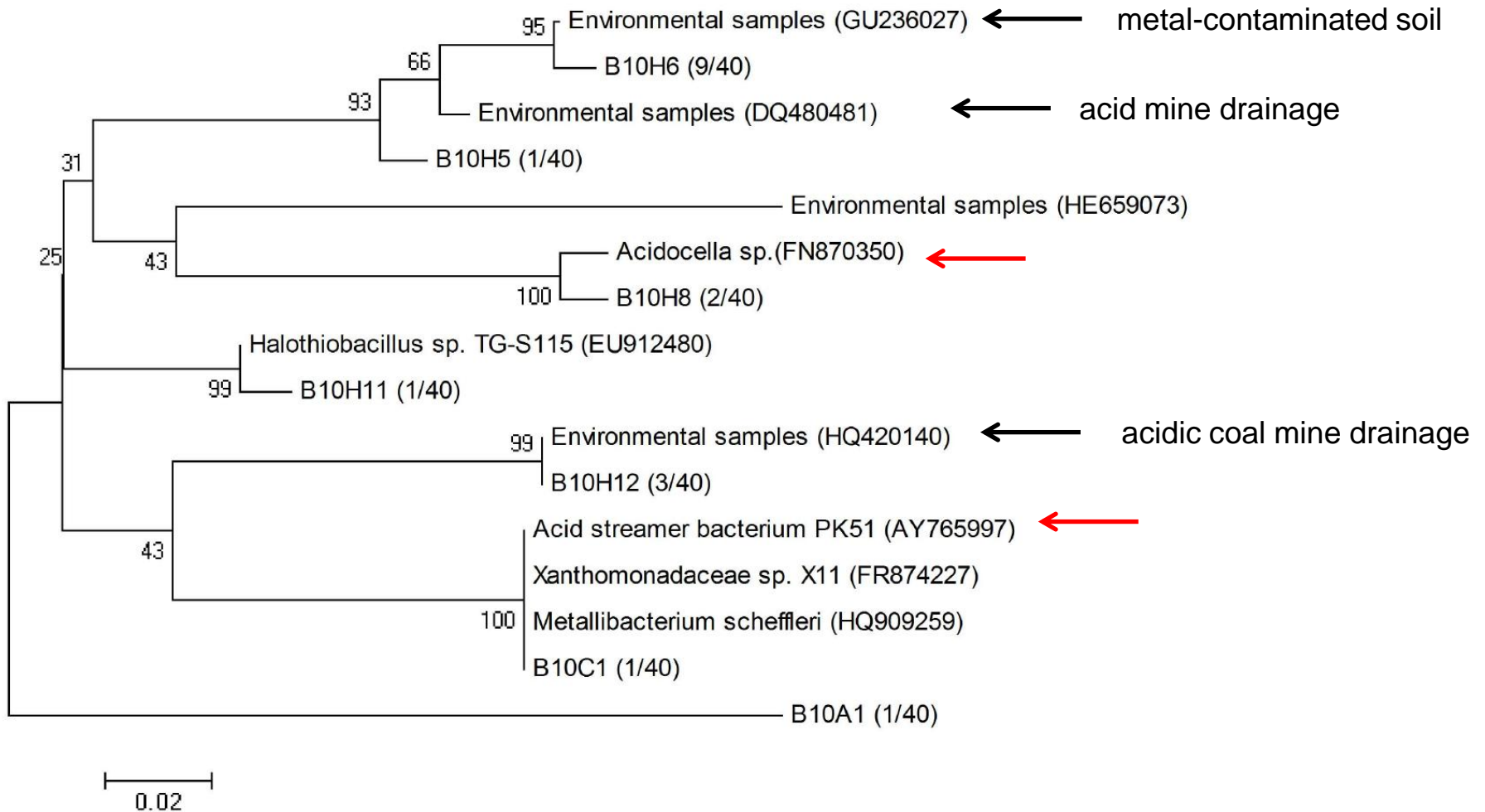
Enrichment on tetrathionate



Metal & Acid Release



Microorganisms After Bioleaching



Lecture

- *Metal dissolution & acid generation*
- *Microorganisms in acid environments*
- *Microorganisms in PASS & ASS*
- **Conclusions**



Conclusions

- Acidophilic microorganisms are found in ASS
- Both ferrous iron and sulfur oxidizers identified
- Initial experiments suggest acidophiles influence metal release from PASS



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