



UiT The Arctic University of Norway

# The geochemical signature of Cu mineralisation preserved in stream sediments from the Alta-Kvænangen Tectonic Window, Northern Norway

*Master's thesis in Geology, GEO-3900, July 2021*

Johan Bang Hilmo

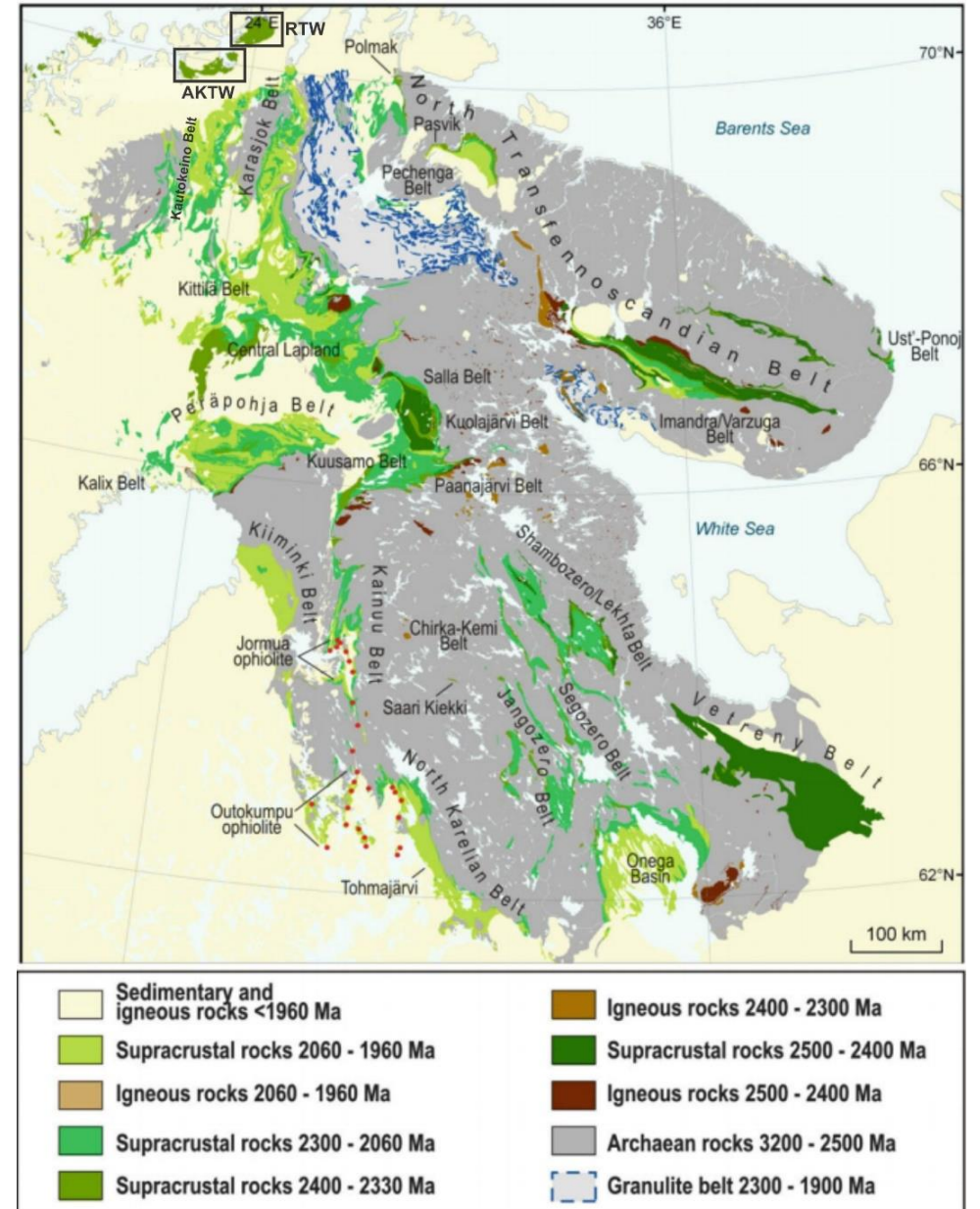
*Main supervisor: Sabina Strmic Palinkaš*

*Co-supervisor: Harald Hansen*



# The Alta-Kvænangen Tectonic Window

- Tectonic window exposed in the Caledonides of Northern Norway
- Hosts several Cu deposits
  - Hosted by both mafic rocks and sedimentary lithologies.



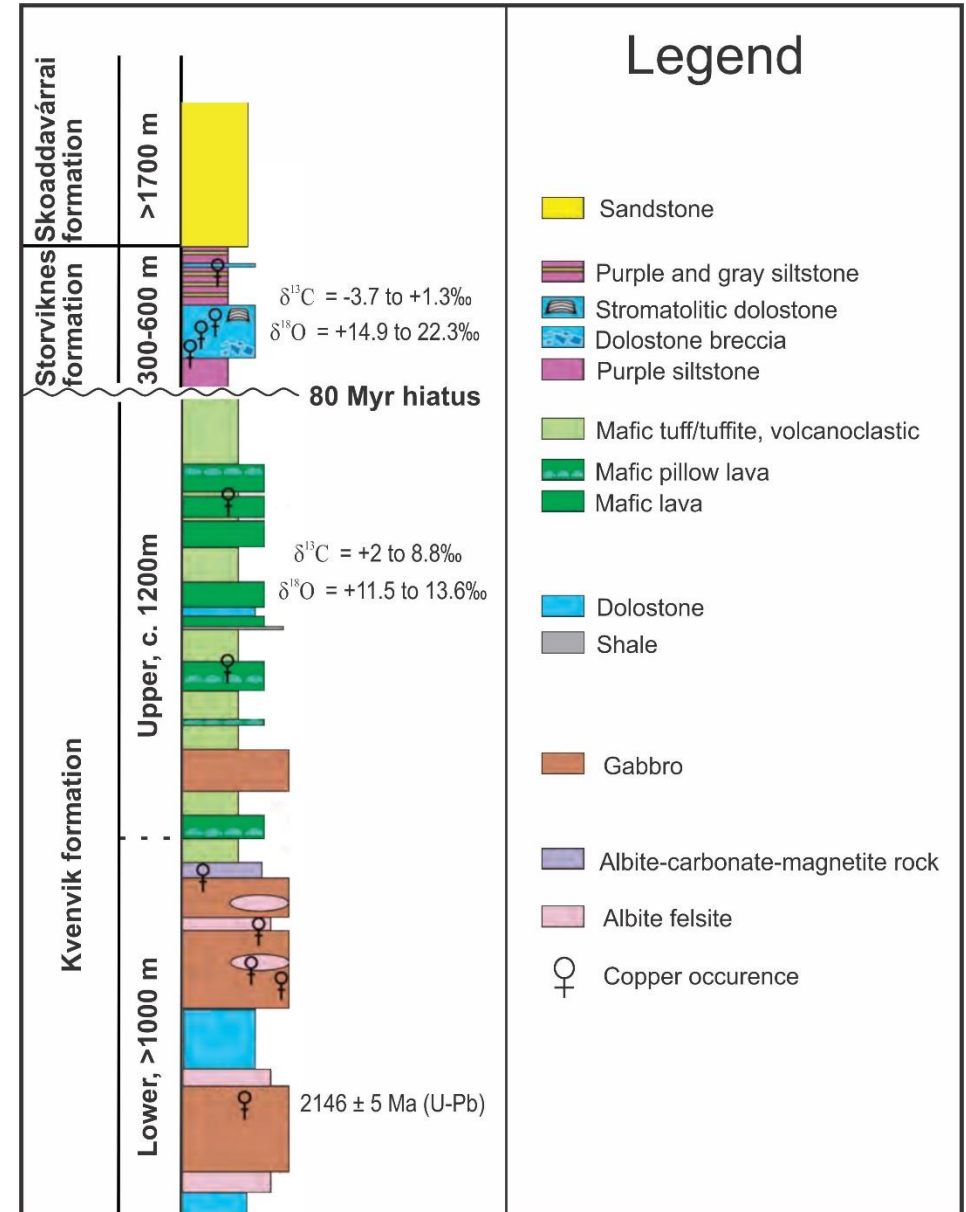
From Melezhik & Hanski (2013)



# Kåfjord, the study area

- Raipas Supergroup
- Carbonate stable isotope compositions ( $\delta^{13}\text{C}$ ,  $\delta^{18}\text{O}$ )
- Mafic rock- and sediment-hosted Cu mineralisation

Stable isotope compositions of carbonates from Simonsen (2021).  
Figure modified from Melezhik et al. (2015)



## Purpose of the study:

Investigate whether the geochemical signature of Cu mineralisation is preserved in stream sediments in the AKTW, and if so to characterise that signature.

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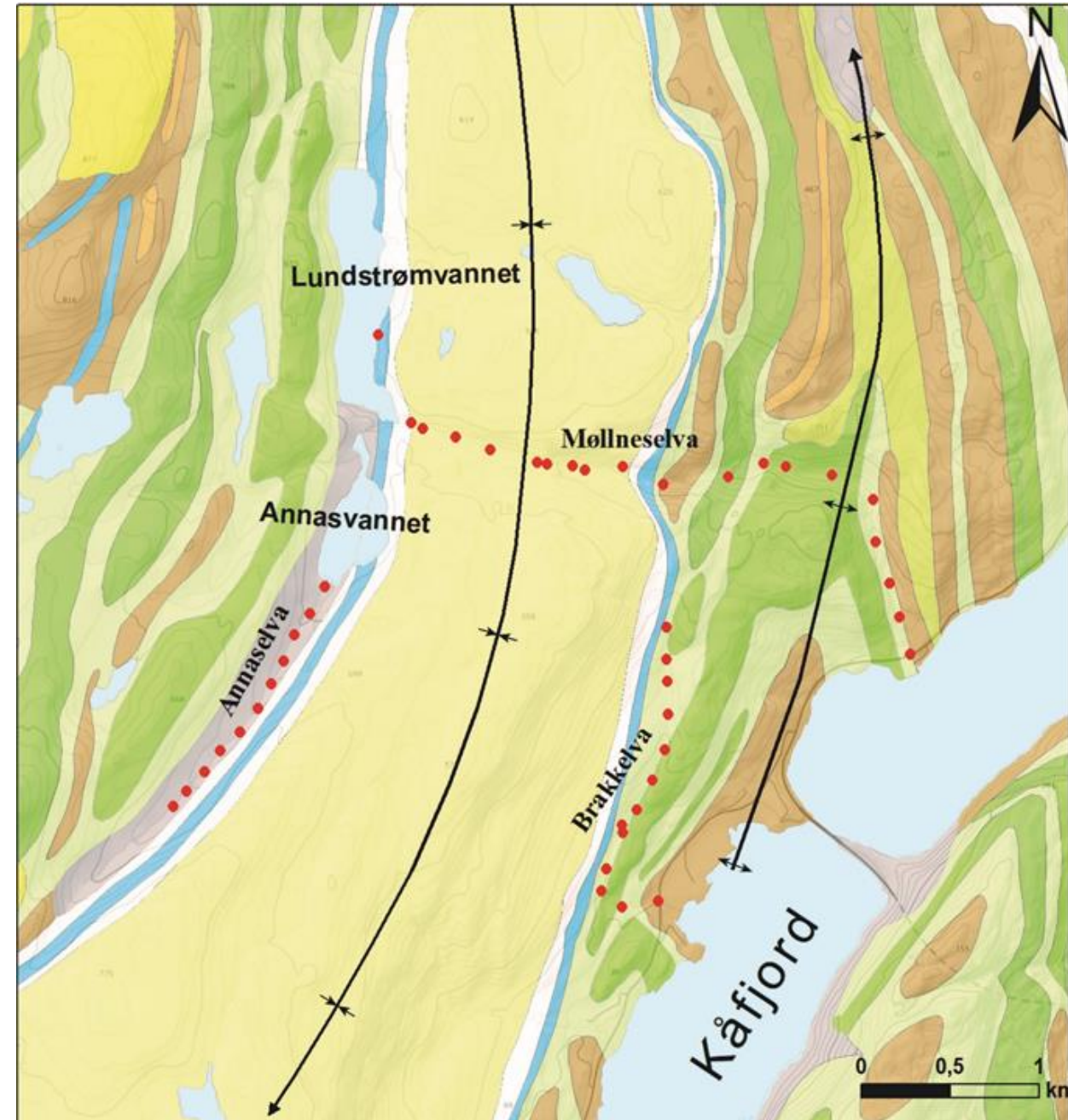
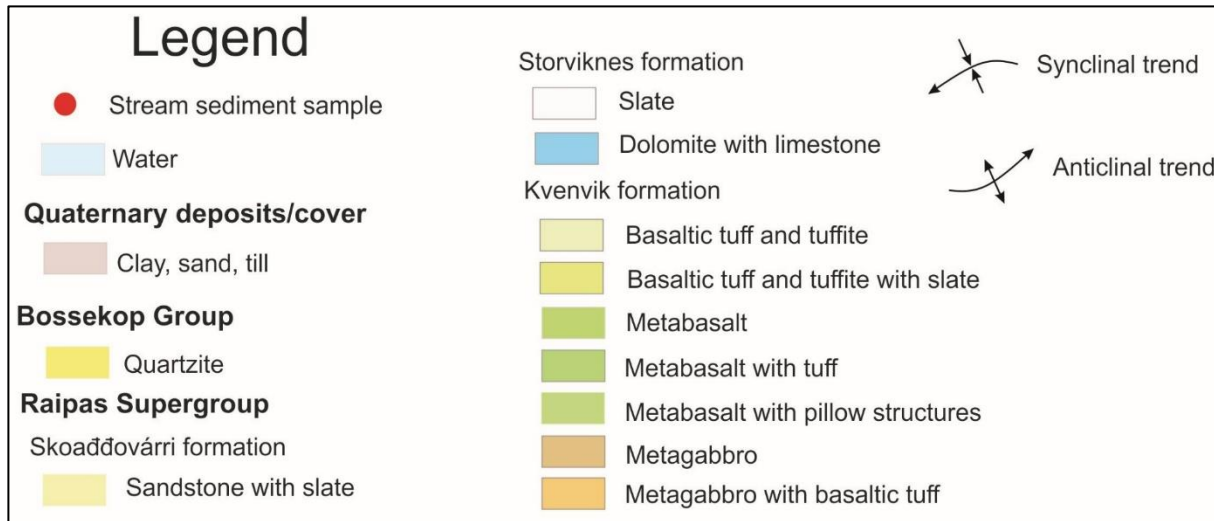
Investigate whether the geochemical signature of Cu mineralisation is preserved in stream sediments in the AKTW, and if so to characterise that signature.

- Bulk chemistry of stream sediments
- Geochemical characteristics of heavy minerals
- Stable isotope composition of carbonates

# Sampling of stream sediments

44 stream sediment samples:

1. Sediment-hosted Cu mineralisation
2. Mafic rock-hosted Cu mineralisation
3. Both the sediment- and mafic rock-hosted Cu mineralisation



Base map modified after *The Geological Survey of Norway (2021)* with structures from *Bergh & Torske (1988)*

## Sieved fractions, additional preparation steps and methods

Fraction ( $\mu\text{m}$ )	Additional preparation	Method
<63		<b>Bulk chemistry</b>
63-125		
125-250	Magnetic separation Picking of minerals	<b>Bulk chemistry</b> SEM-EDS LA-ICP-MS Stable isotope composition of carbonates
250-1000		
>1000		



# Sieved fractions, additional preparation steps and methods

Fraction ( $\mu\text{m}$ )	Additional preparation
<63	
63-125	
125-250	<b>Magnetic separation</b> Picking of minerals
250-1000	
>1000	

- Minerals with different magnetic susceptibilities separated.
- Magnetite and Pyrrhotite extracted using hand held magnet.
- A magnetic separator used for the rest.

	Total range (Amps.)	Best range (Amps.)
Mag	<,01-,05	,01
Hem	,025-,50	,10-,30
Po	<,01-,70	,01-,05
Bn	,30-1,10	,60-,90
Ccp	,30-1,70	,70-1,30
Cct	>1,7	>1,7
Py	>1,7	>1,7
Cal, Dol, Mgs	>1,7	>1,7

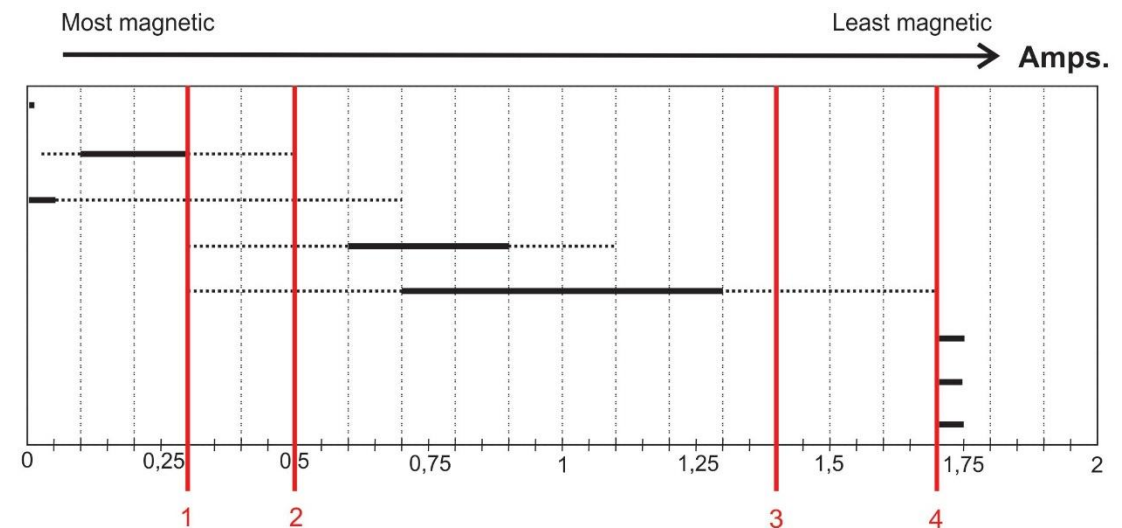


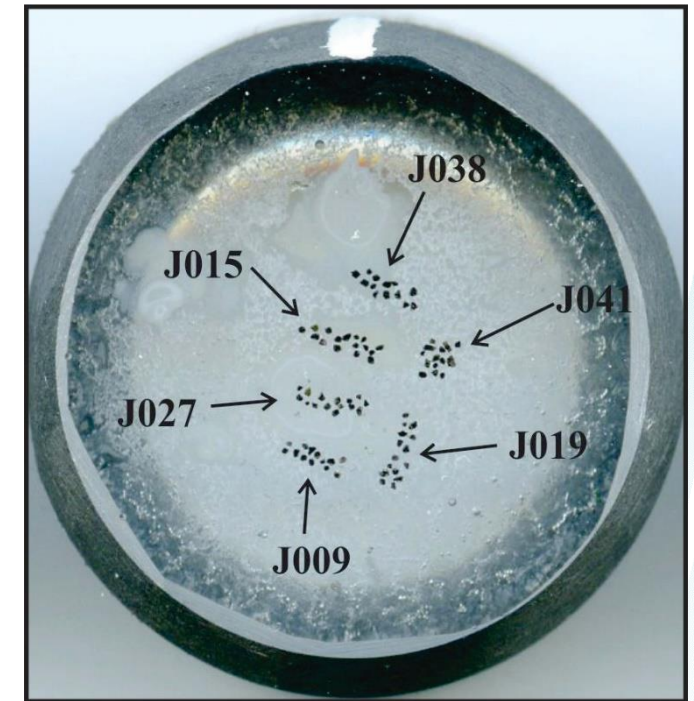
Figure modified after Rosenblum & Brownfield (2000).



# Sieved fractions, additional preparation steps and methods

Fraction ( $\mu\text{m}$ )	Additional preparation
<63	
63-125	
125-250	Magnetic separation <b>Picking of minerals</b>
250-1000	
>1000	

- Hand-picking of minerals from magnetically separated fractions.
  - No further separation needed (e.g. heavy liquids).



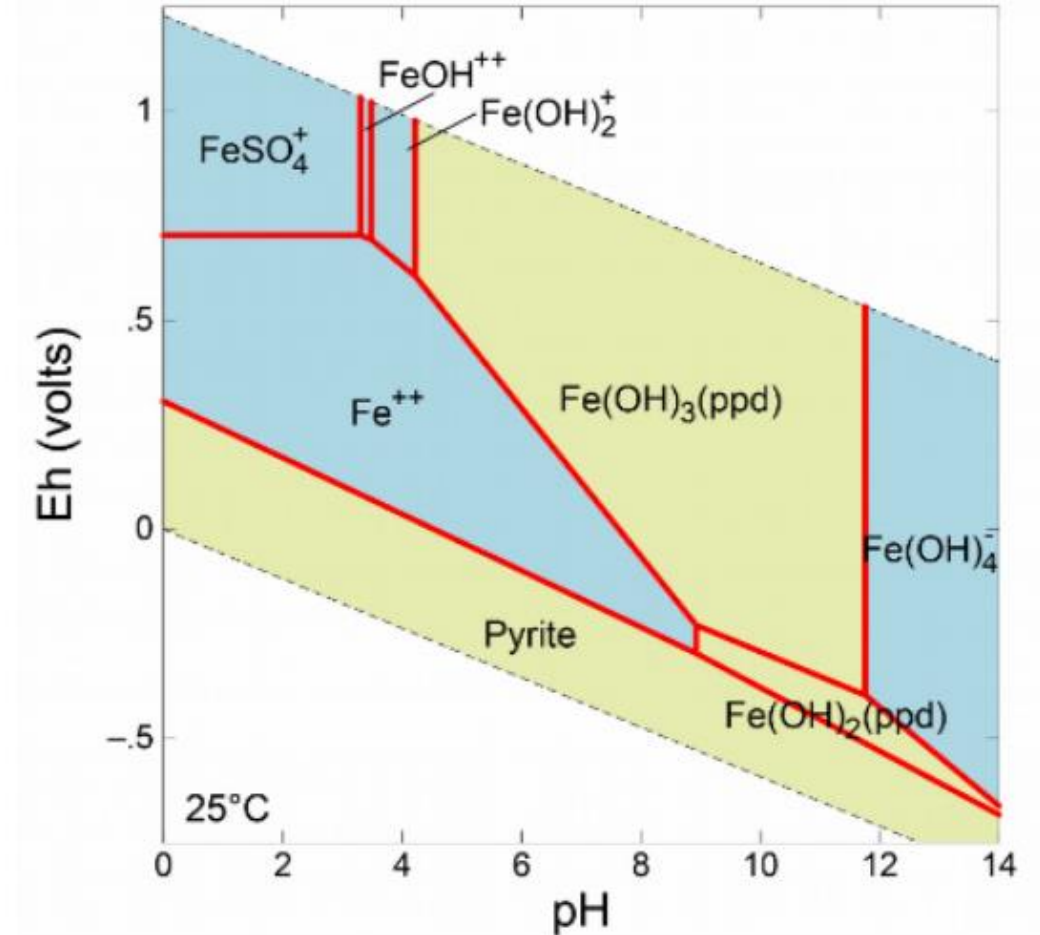
Grains of Mag separated from different samples (J015, J027, ...)

## Sieved fractions, additional preparation steps and methods

Fraction ( $\mu\text{m}$ )	Additional preparation	Method
<63		Bulk chemistry
63-125		
125-250	Magnetic separation Picking of minerals	Bulk chemistry <b>SEM-EDS</b> <b>LA-ICP-MS</b> <b>Stable isotope composition of carbonates</b>
250-1000		
>1000		

## Physicochemical properties (Eh, pH)

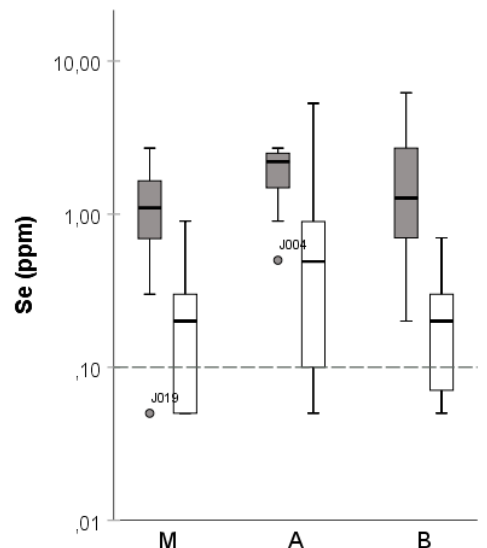
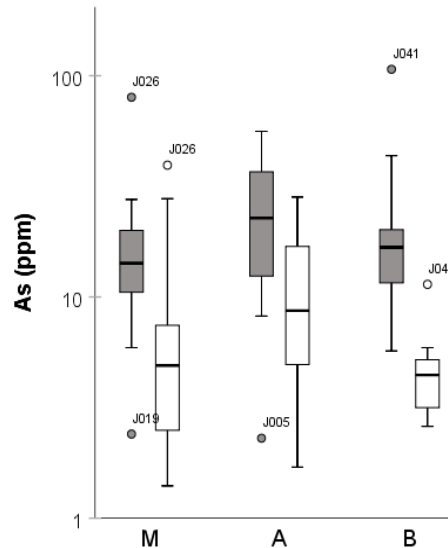
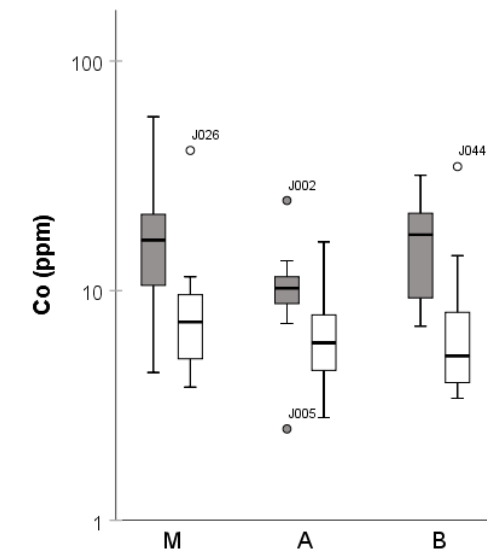
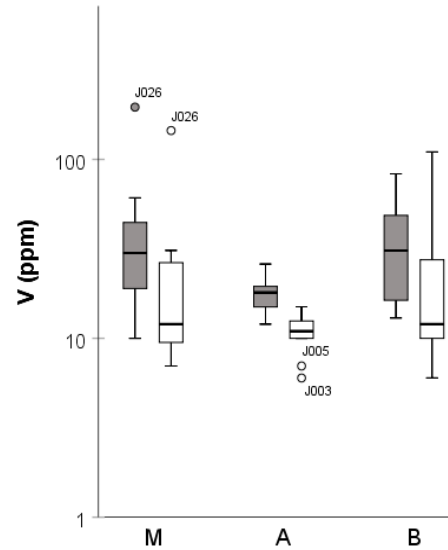
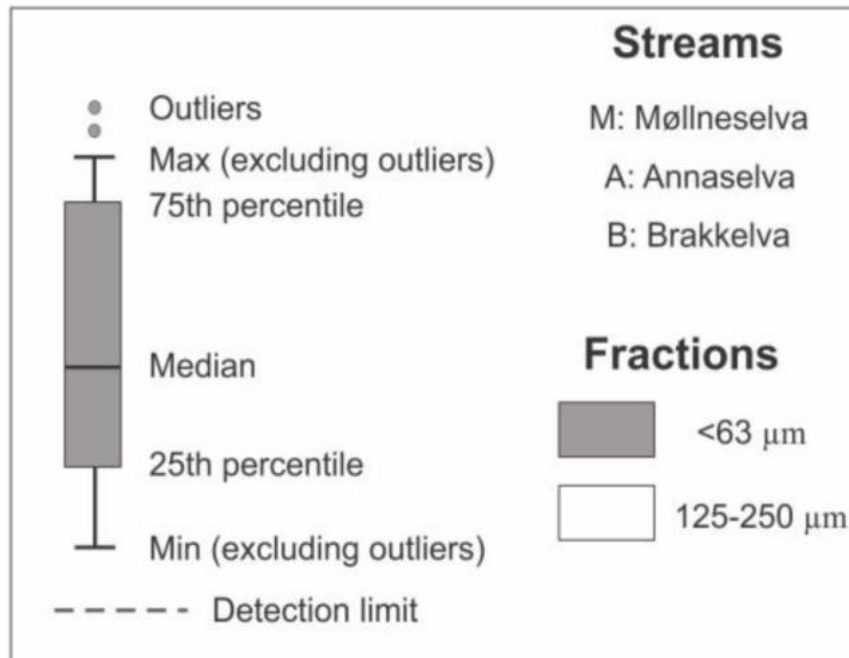
- Redox potential and pH measured in the pore water of 8 samples.
- Eh, pH and the solubility of minerals in surficial water.



From Rose (2010)

# Bulk chemistry of stream sediments

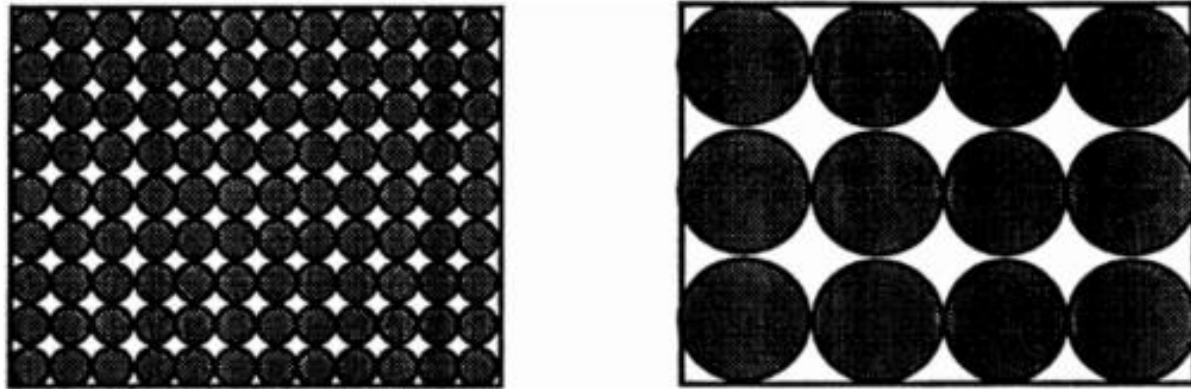
- <math><63 \mu\text{m}</math> versus <math>125\text{-}250 \mu\text{m}</math> fraction.



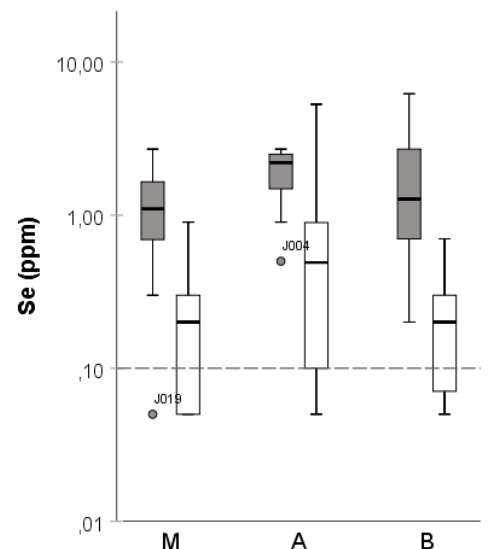
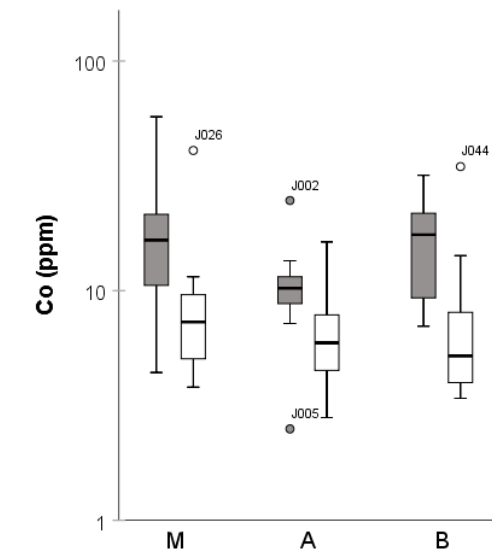
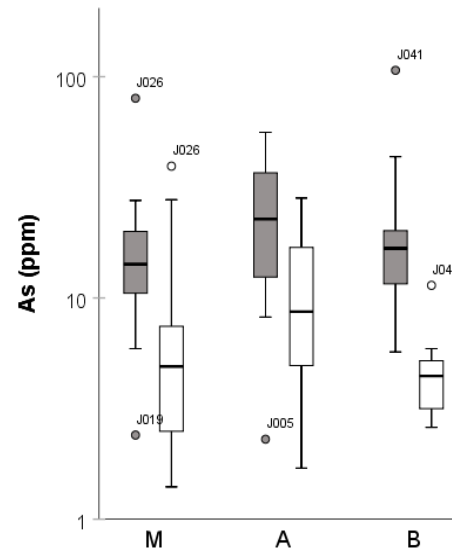
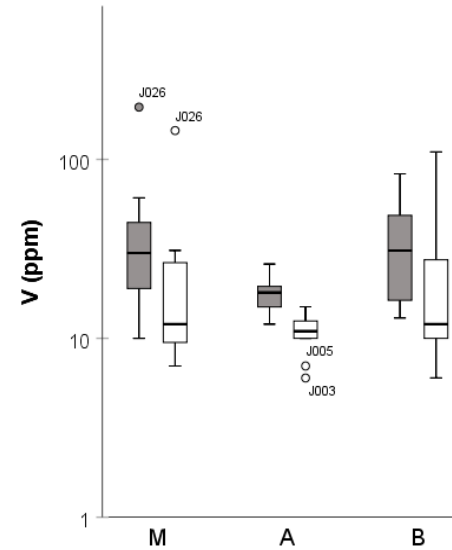


# Bulk chemistry of stream sediments

- $<63 \mu\text{m}$  versus  $125\text{-}250 \mu\text{m}$  fraction.
- Accumulation of dissolved elements on the surface of grains.

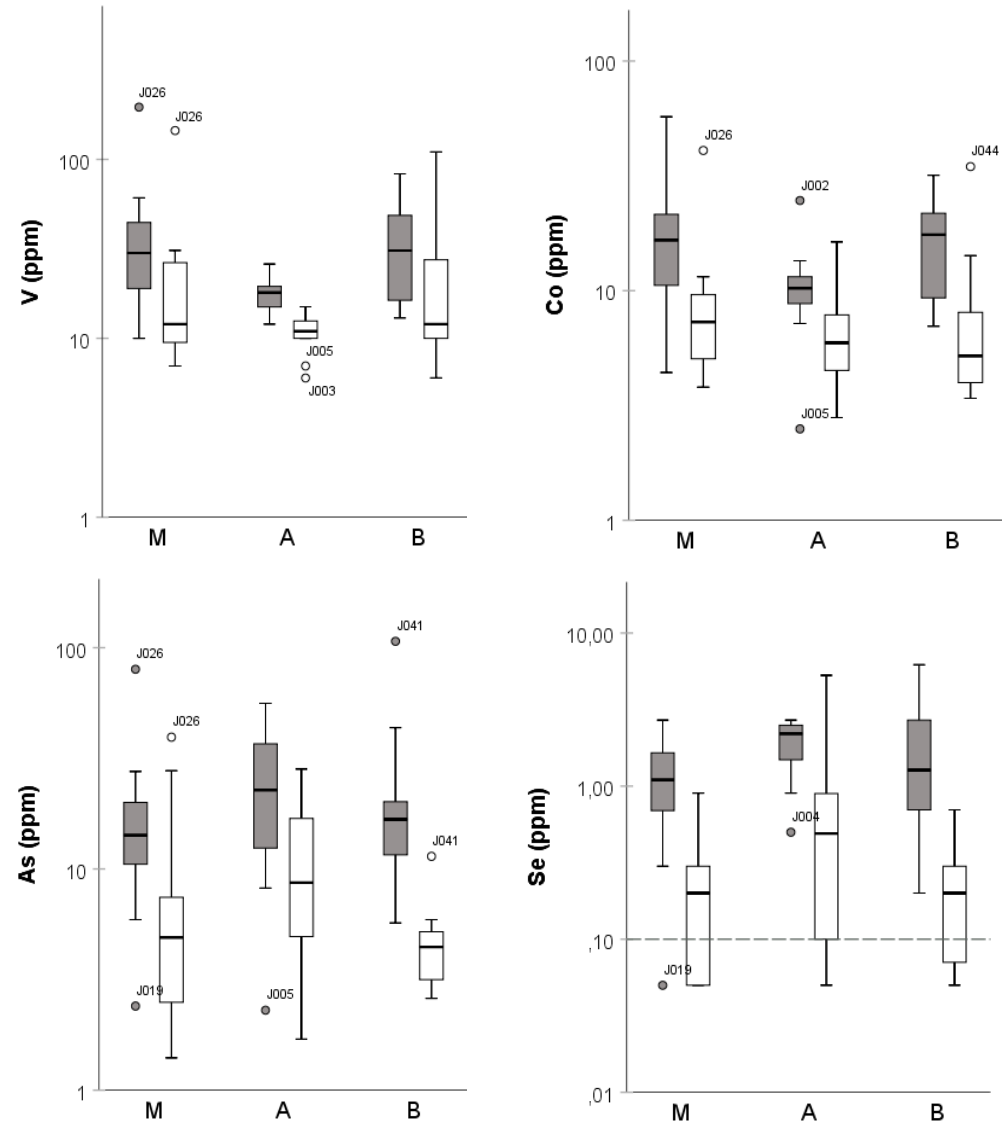


Schematic illustration of how the surface area of grains may vary for a given mass of sediments. From Horowitz (1991).



# Bulk chemistry of stream sediments

- <math><63 \mu\text{m}</math> versus <math>125\text{-}250 \mu\text{m}</math> fraction.
- Accumulation of dissolved elements on the surface of grains.
- Bulk chemistry of sediments from different streams.
  - Møllneselva (M) and Brakkelva (B) drain mafic rocks.
  - Annaselva (A) drain complex mineralisation (e.g. Se-rich sulphide phases) hosted by sedimentary lithologies.

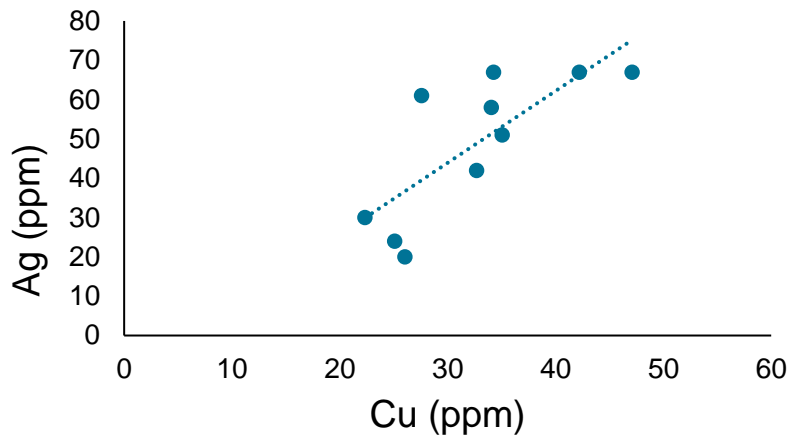


# Pathfinder elements and correlation

- Cu correlates with chalcophile elements (e.g Ag, Zn, Mo).

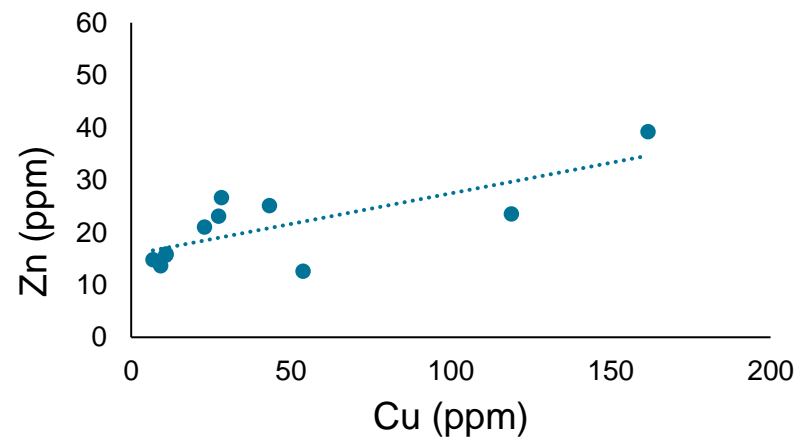
## Draining carbonate lithologies

<63  $\mu\text{m}$  fraction, Annaselva stream



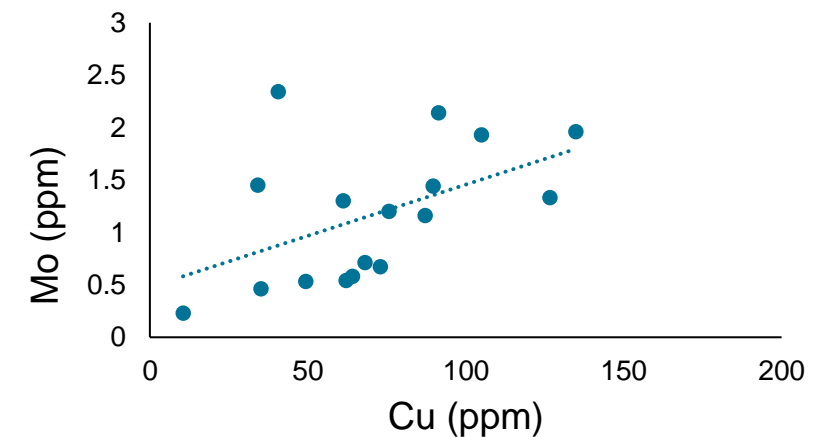
## Draining mafic lithologies

125-250  $\mu\text{m}$  fraction, Brakkelva stream



## Draining all lithologies

<63  $\mu\text{m}$  fraction, Møllneselva stream

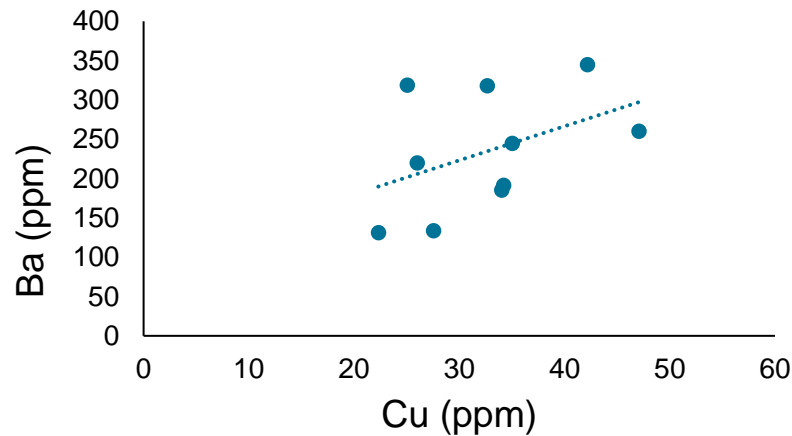


# Pathfinder elements and correlation

- Cu correlates with chalcophile elements (.e.g Ag, Zn, Mo).
- Divalent cations, for example barium, may substitute Ca in  $\text{CaCO}_3$  ( $\text{Ba}^{2+} \leftrightarrow \text{Ca}^{2+}$ ).

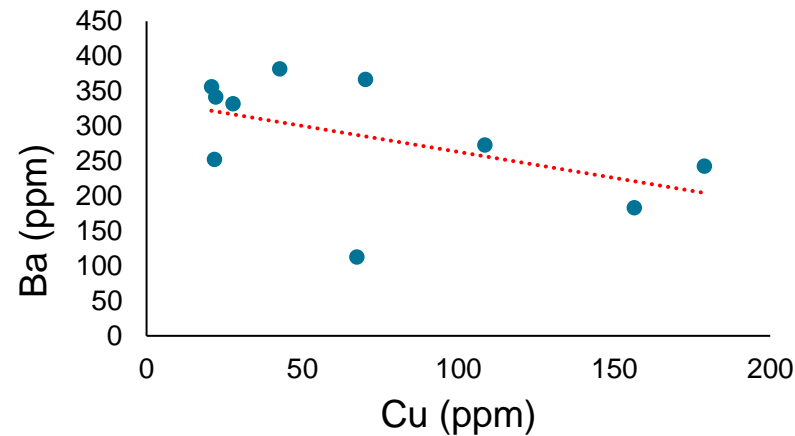
## Draining carbonate lithologies

<63  $\mu\text{m}$  fraction, Annaselva stream



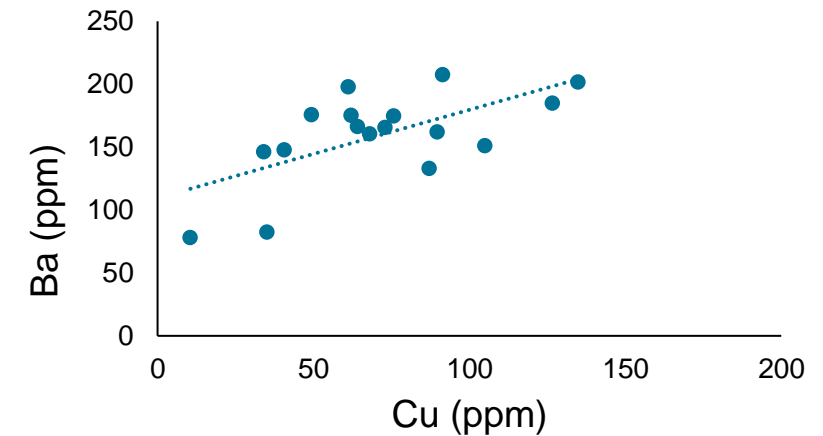
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<63  $\mu\text{m}$  fraction, Brakkelva stream



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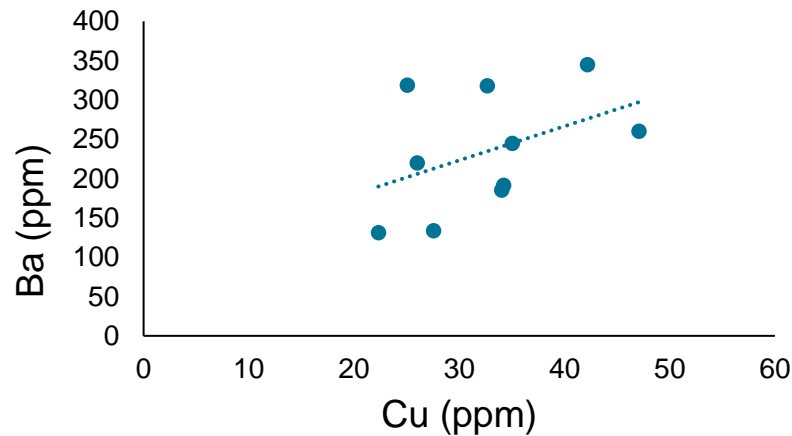
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**Does the correlation reflect the Cu mineralisation occurring in carbonate-rich lithologies?**

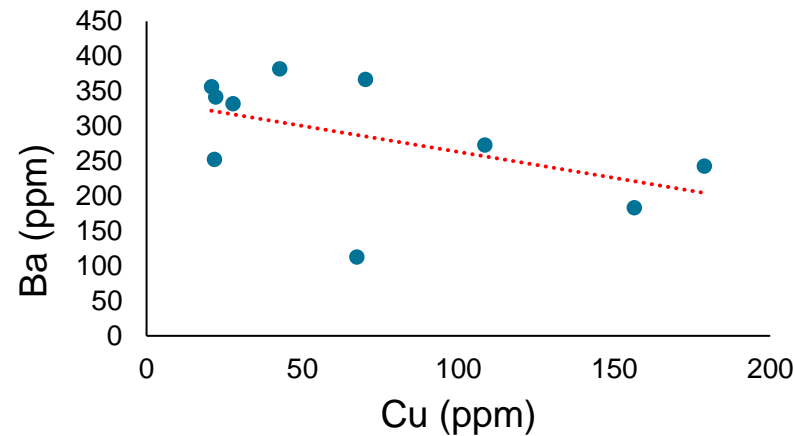
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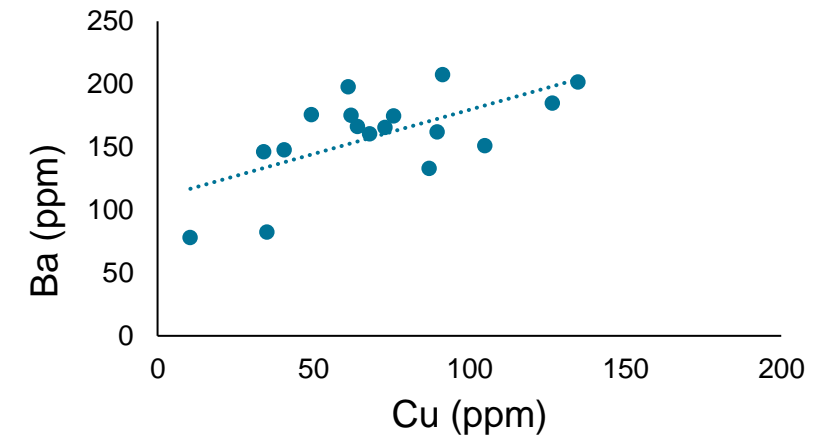
## Draining mafic lithologies

<63  $\mu\text{m}$  fraction, Brakkelva stream



## Draining all lithologies

<63  $\mu\text{m}$  fraction, Møllneselva stream

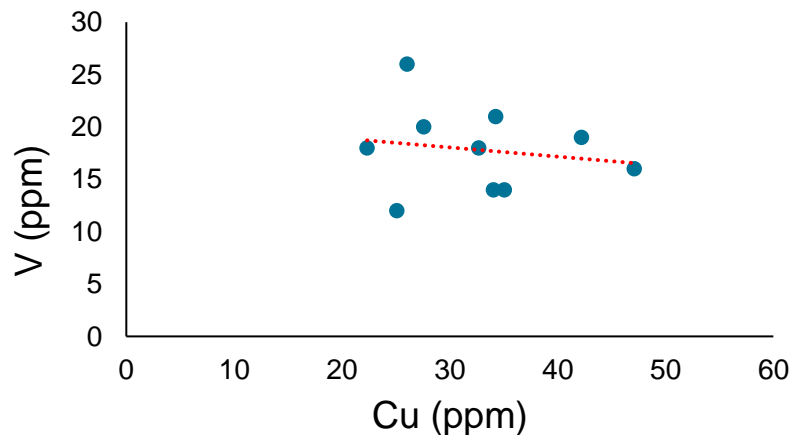


# Pathfinder elements and correlation

- Cu correlates with chalcophile elements (.e.g Ag, Zn, Mo).
- Divalent cations, for example barium, may substitute Ca in  $\text{CaCO}_3$  ( $\text{Ba}^{2+} \leftrightarrow \text{Ca}^{2+}$ ).
- Similarly, the correlation between Cu and siderophile elements (V, Ni, Co) may reflect the mafic host rock lithology.

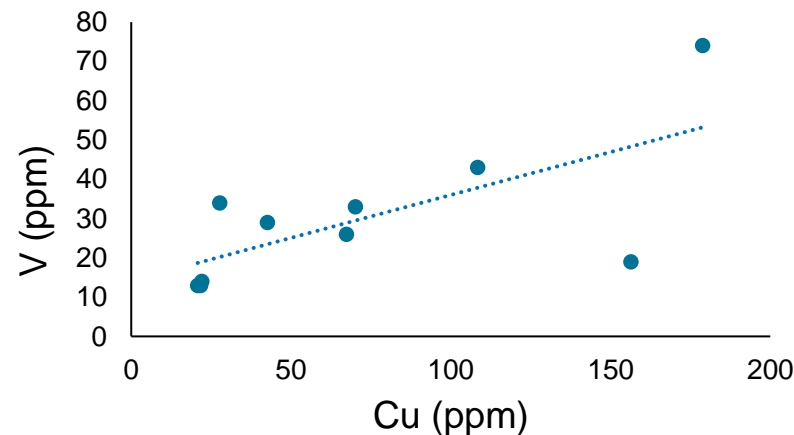
## Draining carbonate lithologies

<63  $\mu\text{m}$  fraction, Annaselva stream



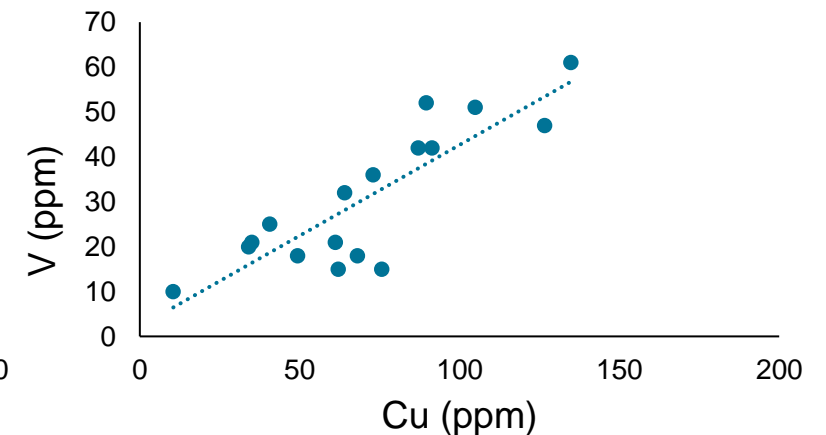
## Draining mafic lithologies

<63  $\mu\text{m}$  fraction, Brakkelva stream



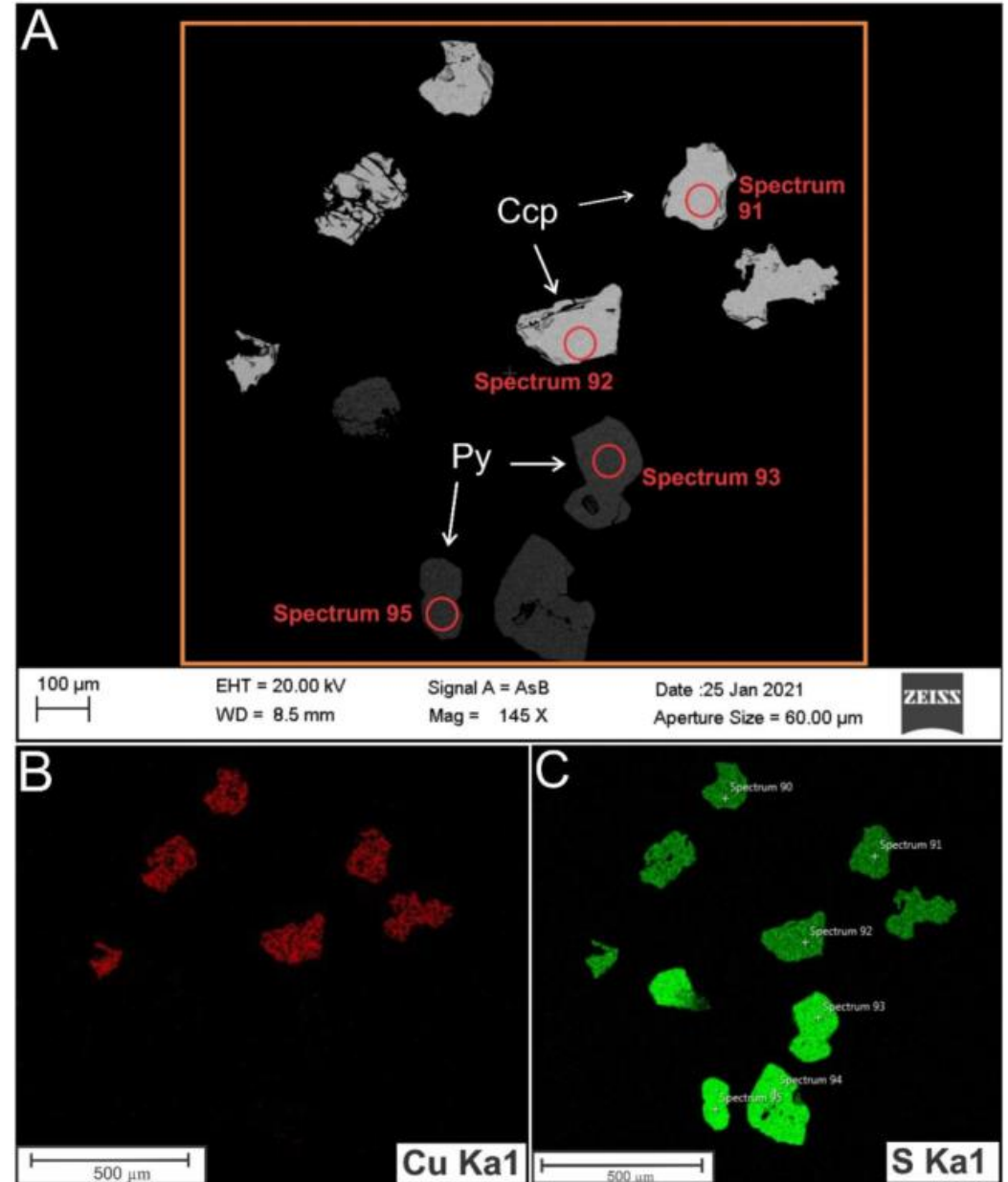
## Draining all lithologies

<63  $\mu\text{m}$  fraction, Møllneselva stream



# SEM-EDS analyses of heavy minerals

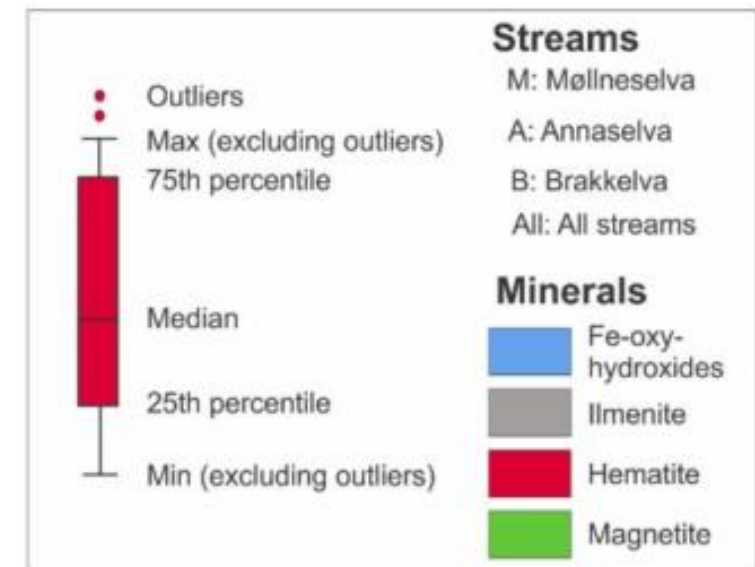
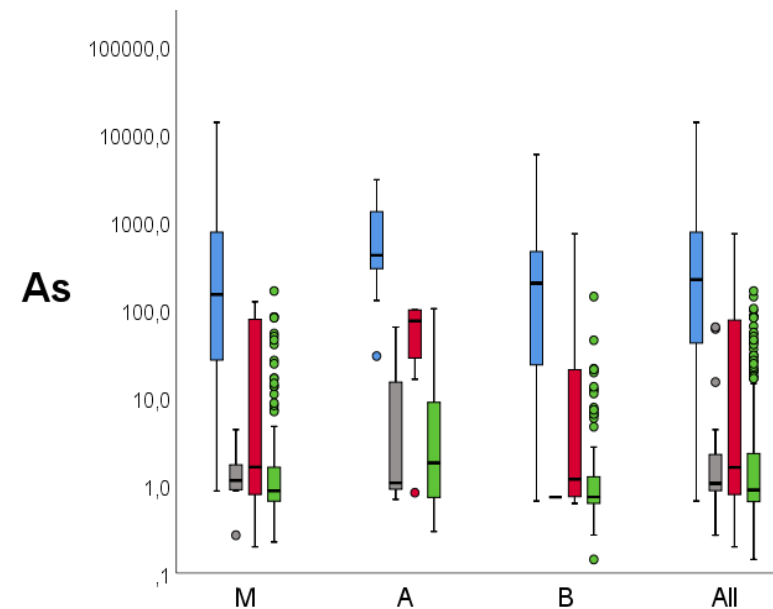
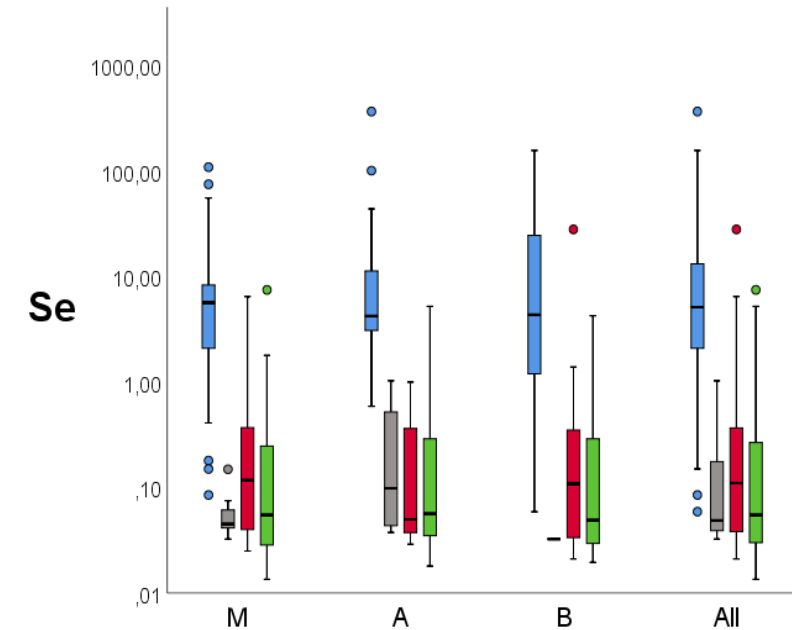
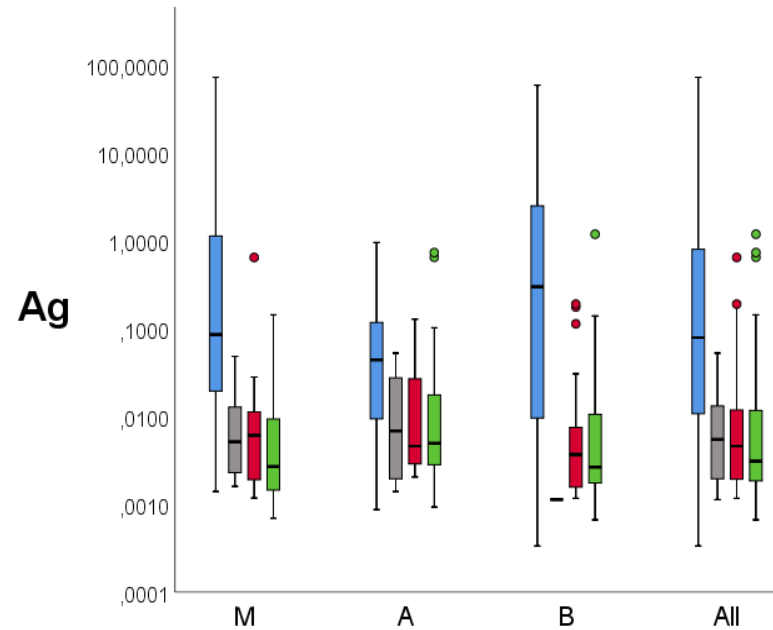
- Confirm the mineralogy of grains.
- Select grains to be analysed by LA-ICP-MS.
- Determine internal standards (wt.% Fe).



# LA-ICP-MS of heavy minerals

- Oxides analysed:

- 76 Fe-oxy-hydroxides
- 18 ilmenite
- 59 hematite
- 351 magnetite

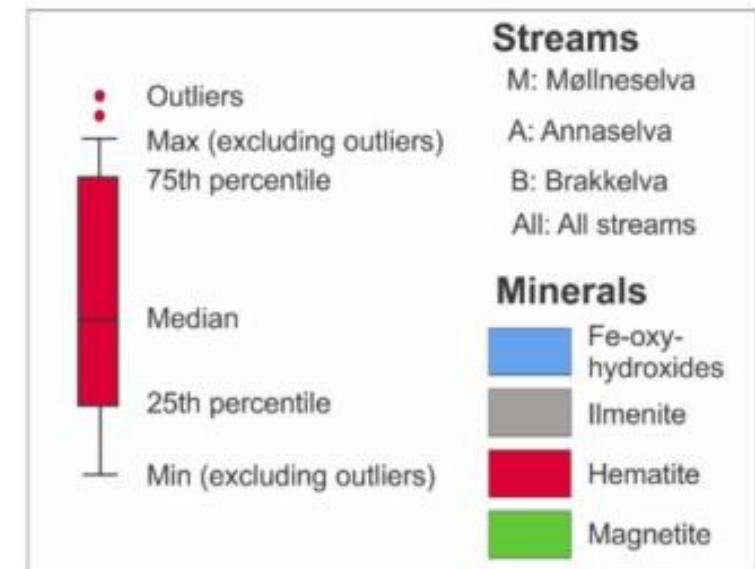
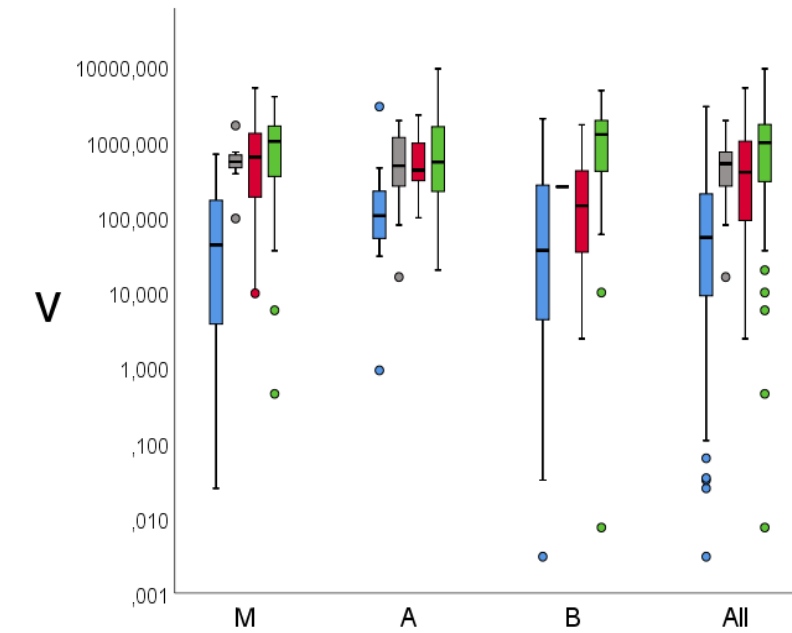
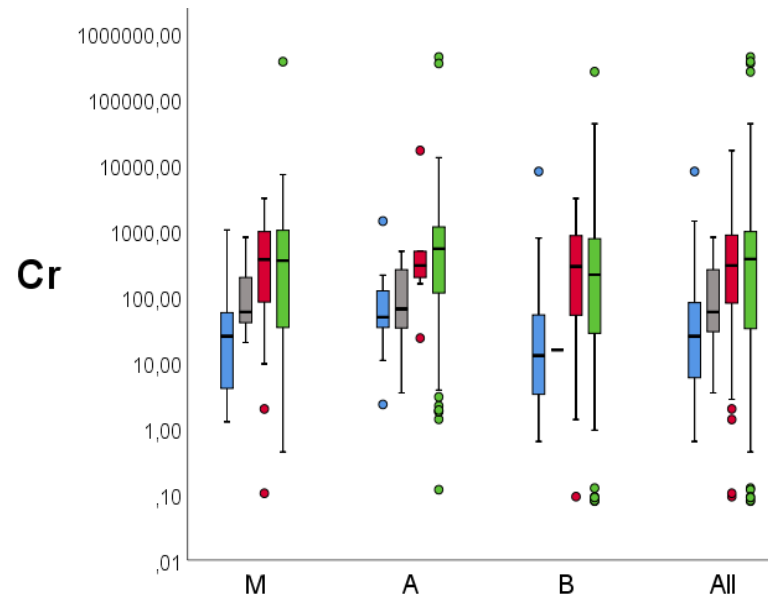
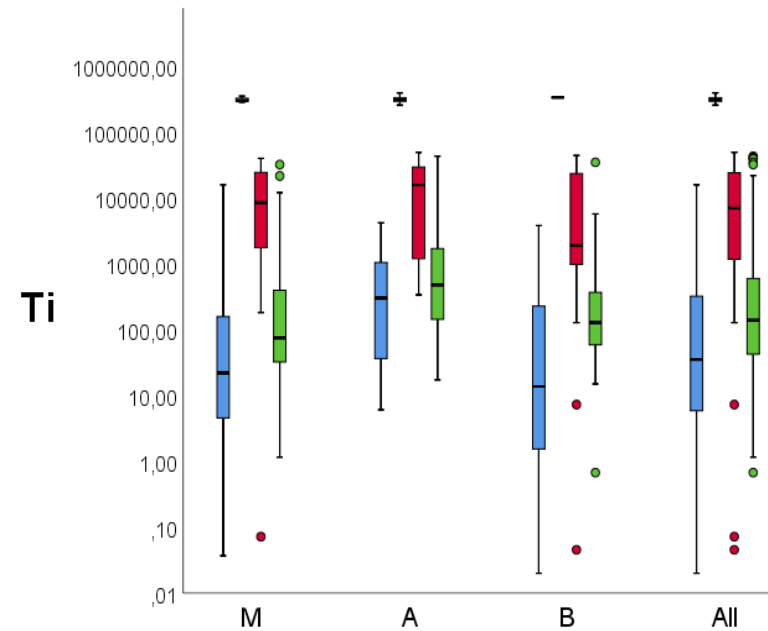




# LA-ICP-MS of heavy minerals

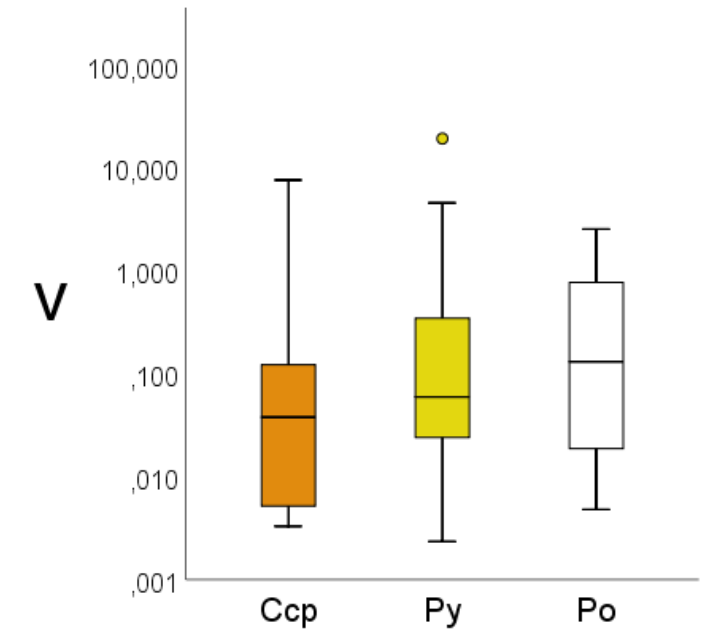
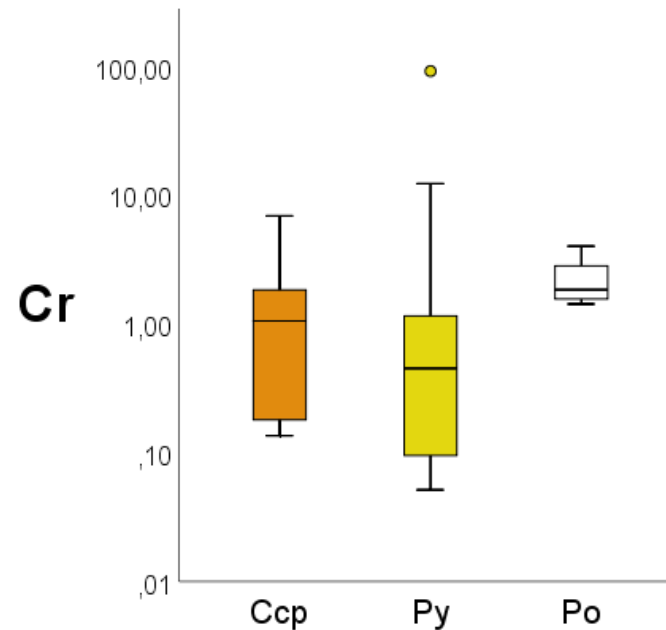
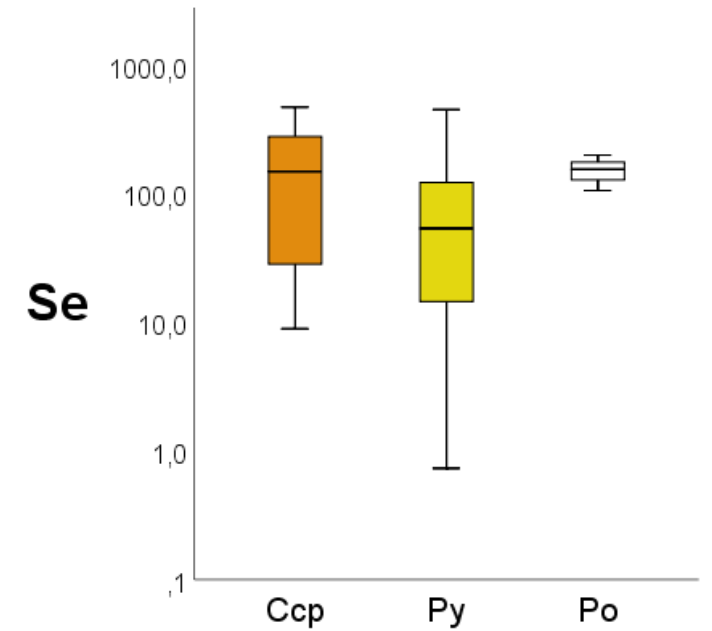
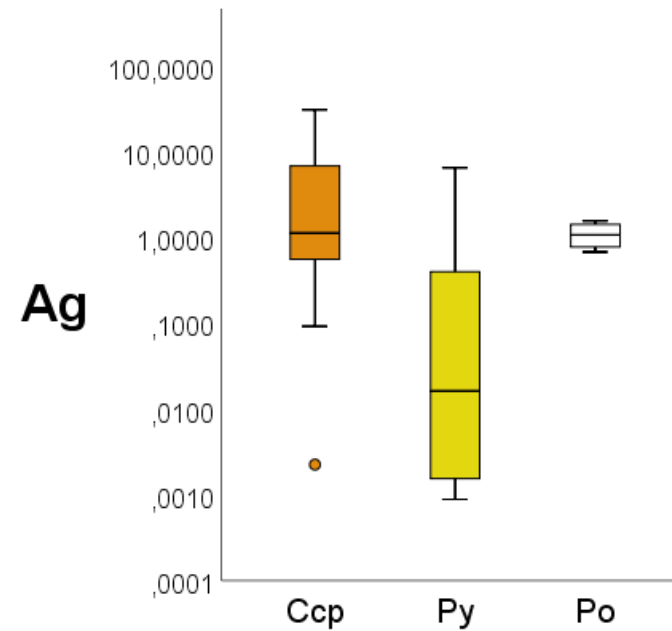
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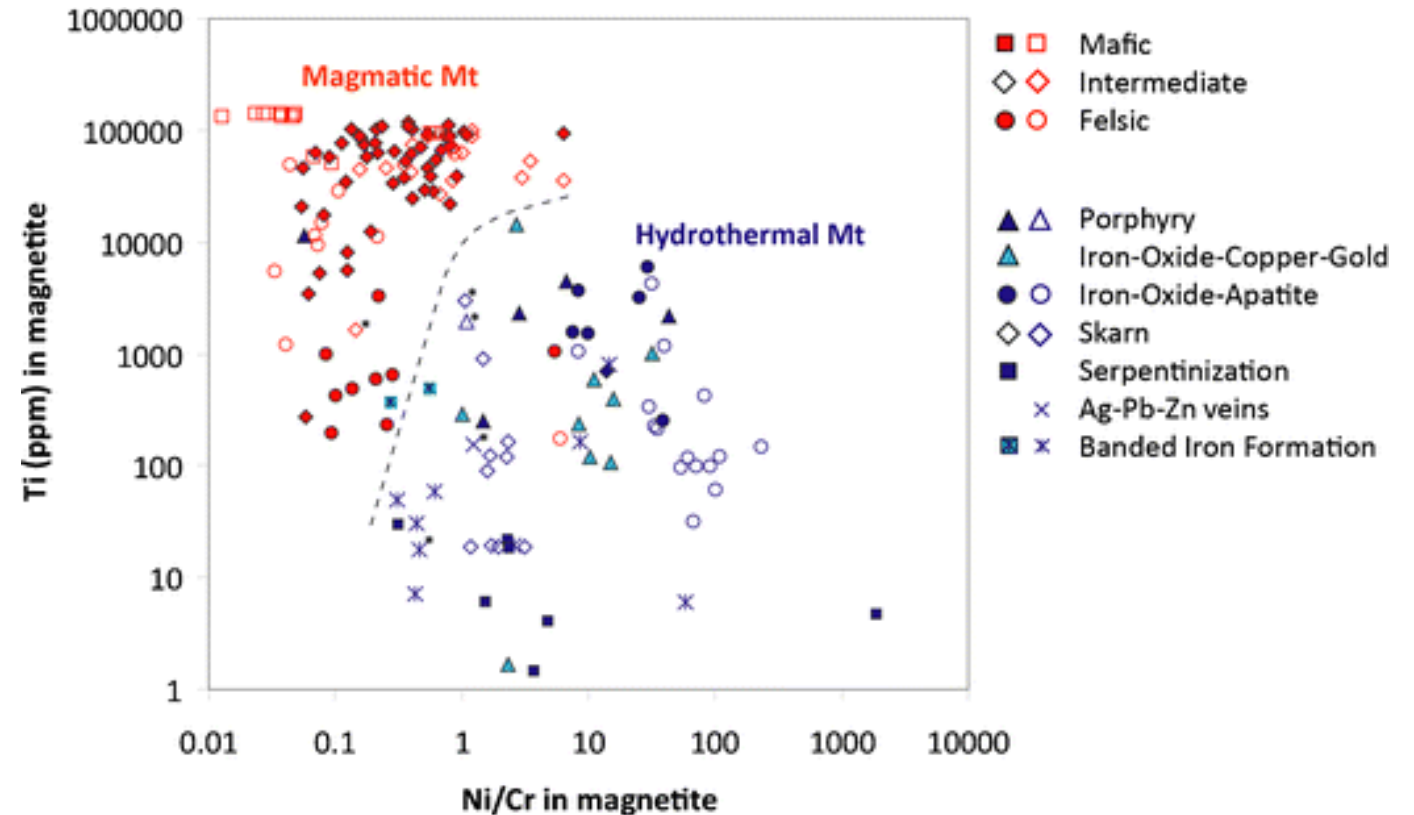
# LA-ICP-MS of heavy minerals

- Sulphides analysed:
  - 32 chalcopyrite
  - 39 pyrite
  - 4 pyrrhotite



# Indicator minerals, LA-ICP-MS of heavy minerals

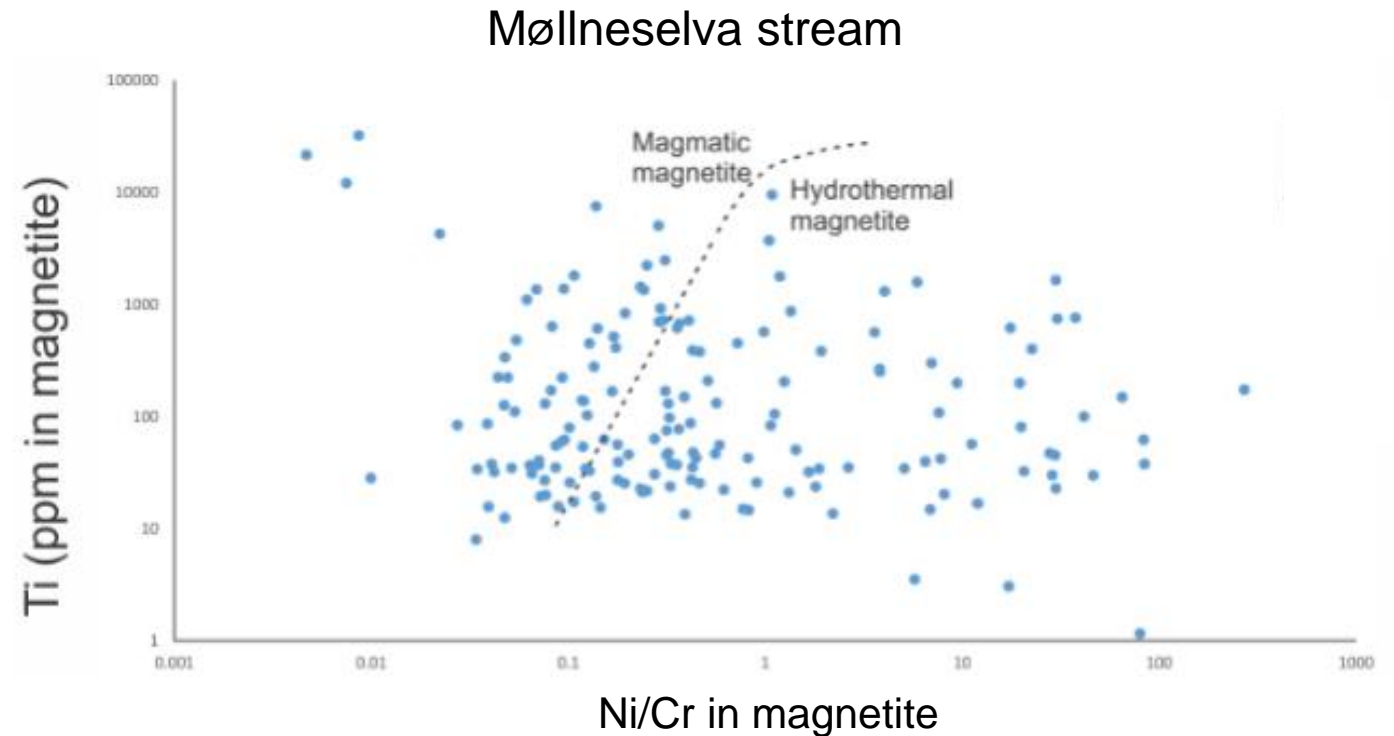
- Composition of minerals can differ with respect to the environment in which they are formed in.
  - Example: Magnetite of magmatic/hydrothermal origin.



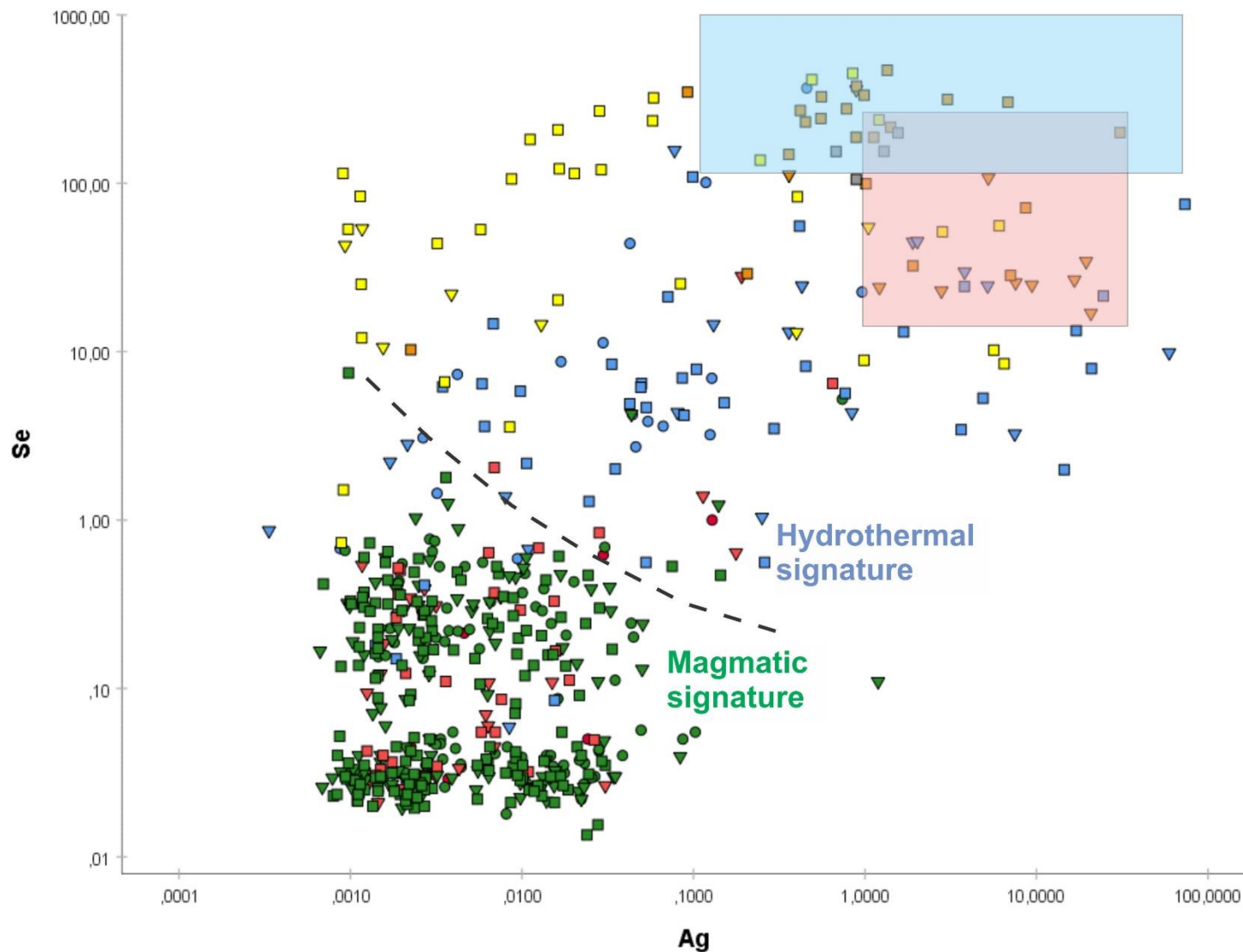
From Dare et al. (2014)

# Indicator minerals, LA-ICP-MS of heavy minerals

- Composition of minerals can differ with respect to the environment in which they are formed in.
  - Example: Magnetite of magmatic/hydrothermal origin.
- Magnetite does not seem to be a reliable indicator mineral in AKTW.



# Indicator minerals, LA-ICP-MS of heavy minerals



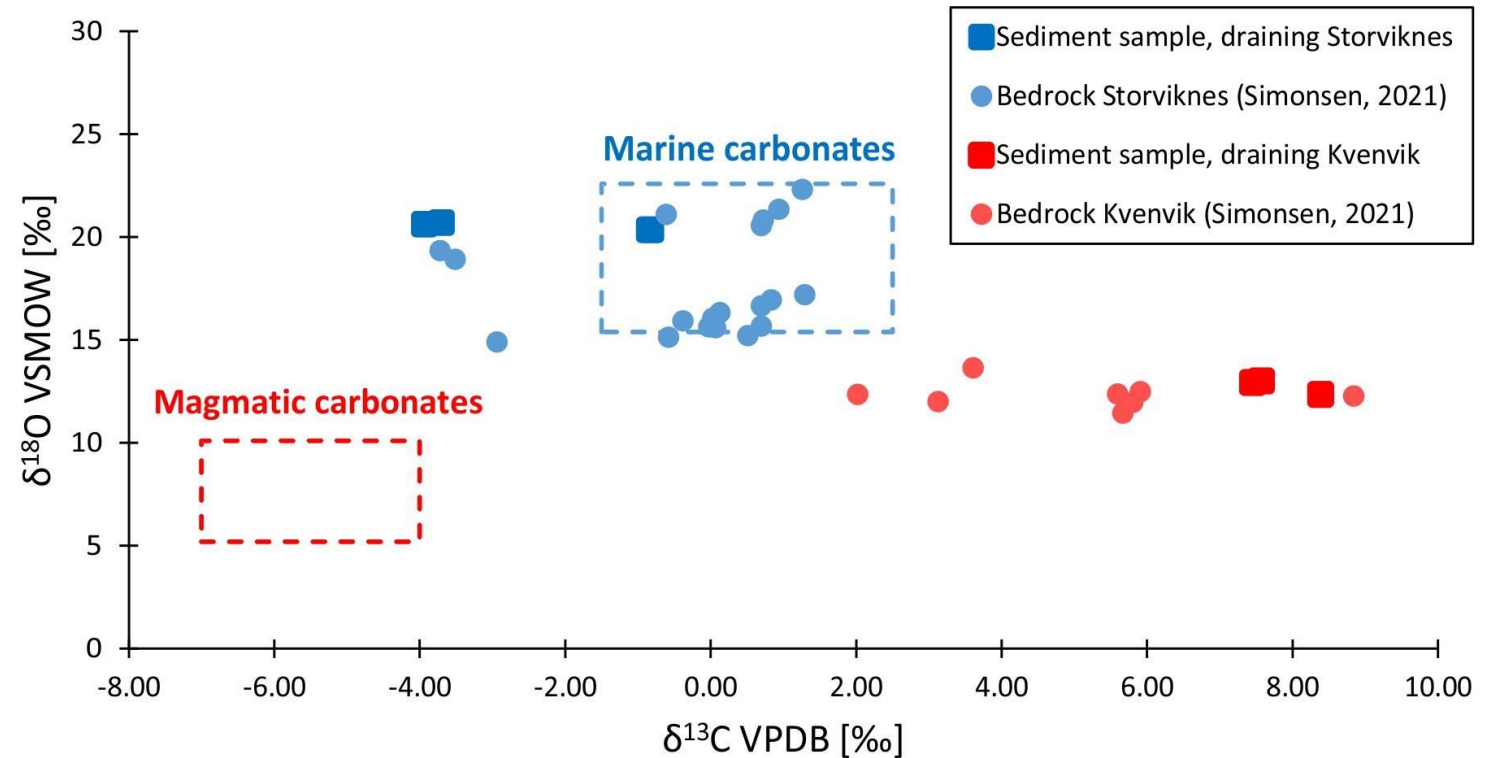
Mineral	Stream		
	Annaselva	Brakkelva	Møllneselva
Fe-oxy-hydroxides	●	▼	■
Hematite	●	▼	■
Magnetite	●	▼	■
Pyrite		▼	■
Chalcopyrite		▼	■
Pyrrhotite			■

**Minor and trace element contents in hardrock samples (Simonsen, 2021)**

Chalcopyrite (Storviknes formation)	■
Chalcopyrite (Kvenvik formation)	■

# Carbonate stable isotopes

- Two groupings:
  - Storviknes: high positive  $\delta^{18}\text{O}$ ,  $\delta^{13}\text{C} \sim 0 \text{ ‰}$ .
  - Kvenvik: Lower, but positive  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  typically +3 to +10 ‰.



*Reference isotopic compositions of magmatic carbonates: Stakes & O'Neil (1982) and marine carbonates: Veizer & Hoefs (1976).*



# Conclusions

- The  $<63 \mu\text{m}$  fraction is enriched in most of analysed elements.
- A different bulk chemical signature is displayed in sediments from each of the streams.
  - Chalcophile elements (hydrothermal signature).
  - Divalent cations (carbonate-rich lithologies).
  - Siderophile elements (mafic rocks hosting the Cu mineralization).
- Magnetic separation + hand picking of heavy minerals is efficient.
- Hydrothermal Cu mineralisation: high Ag and Se.
  - Signature of Fe-oxy-hydroxides resembles the hydrothermal signature of sulphides. Indicator mineral?
- Isotopic signature of carbonates is preserved in stream sediments.

# Thank you!



Photo: Sabina Strmic Palinkaš